

The Origin of the Elements

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Outline:

1. What is an element?
2. Elements are made by nucleosynthesis in stars
3. How do we know the abundances of elements in stars:
 - o Stellar spectra
 - o Stellar dust particles in meteorites
4. Calculating the evolution of stars and the making of the elements
5. Scientists paradise: new data, lots of it!
6. Conclusions

What is an element

Basic building blocks of everything we see (breeze!) around us:

- o Air: nitrogen (N, 78.09%), oxygen (O, 20.95%), argon (Ar, 0.93), carbon dioxide (CO_2 , 0.03), trace gases (0.003%)
- o Metals:
 - steel: alloy of iron (Fe) and less than 2% carbon (C)
 - bronze, brass: alloy of Tin (Sn) and copper (Cu)
- o Glass: mainly sand: silicon dioxide (SiO_2)
- o Water: hydrogen and oxygen: H_2O
- o The Sun (and most other stars): hydrogen (H, 0.76%), helium (He, 0.22%), and a little bit of everything else ...

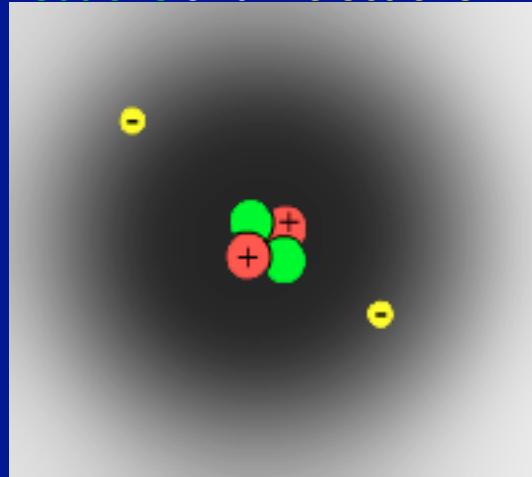
Everything else are 105 elements.

What is an element

Material that consists of atoms with the same number of protons in the nucleus.

Helium atom model:

Nucleus with 2 protons, 2 neutrons and 2 electrons



A nucleus consists of protons and neutrons.
Nuclei with the same number of protons are the same element. Elements have different isotopes with different neutron numbers.

<-- this is ${}^4\text{He}$

This:  is ${}^3\text{He}$

More elements and their proton/charge number:

C - 6, N - 7, O - 8, Ne (neon) - 10, Si - 14, Fe - 26, Pb (lead) 82

The stable isotopes of C: ${}^{12}\text{C}$ and ${}^{13}\text{C}$

Bring order to this mess ...

The periodic table

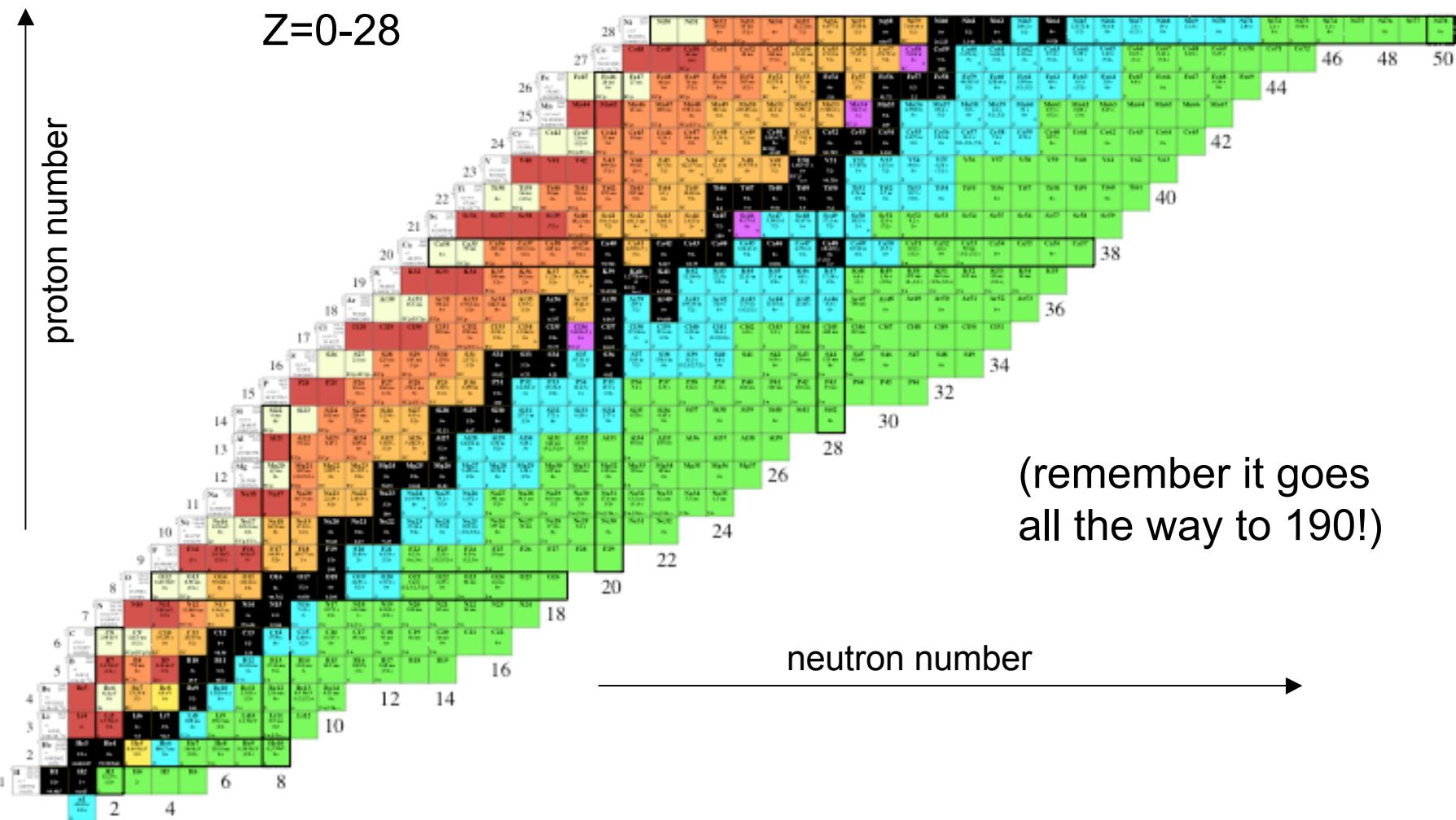
Order by number of electrons -> chemistry

Not suitable for nuclear (astro-) physics!!!

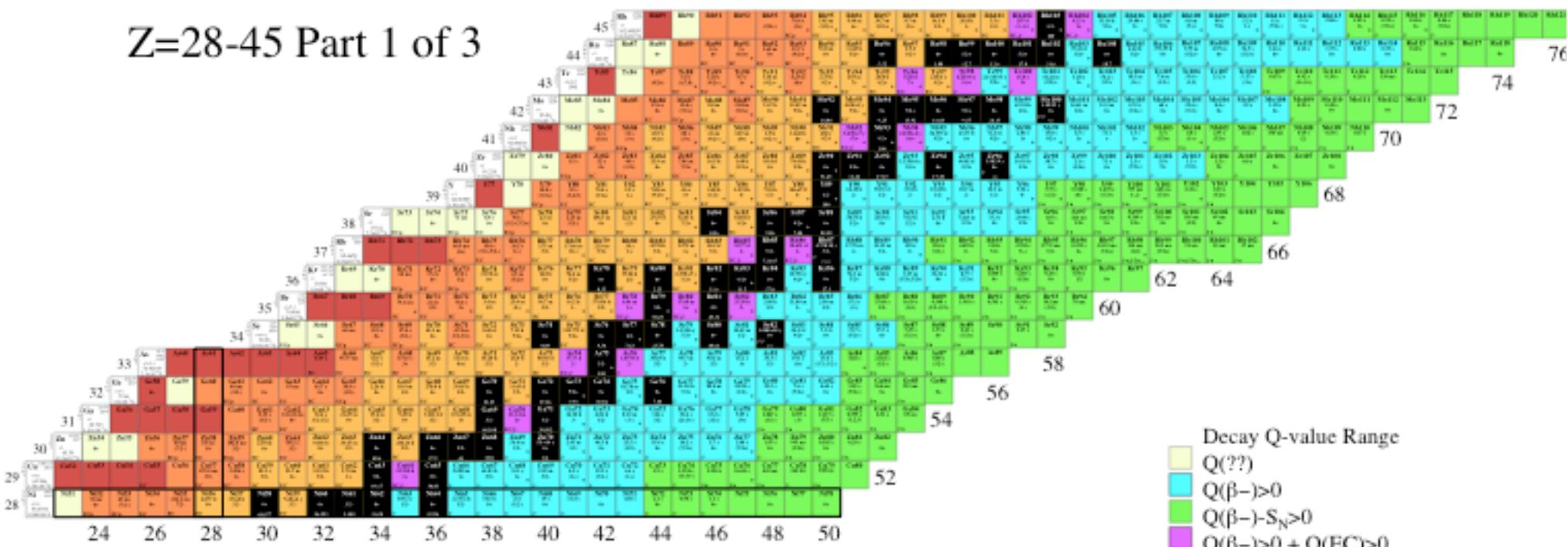
The periodic table of the elements																												
1A	2A	3A	4A	5A	6A	7A	8	1B	2B	3B	4B	5B	6B	7B	0													
1 H								2 He																				
2 Li	3 Be							5 B	6 C	7 N	8 O	9 F	10 Ne															
3 Na	12 Mg							13 Al	14 Si	15 P	16 S	17 Cl	18 Ar															
4 K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	31 Ga	Ge	As	Se	Br	Kr											
5 Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	50 Sn	51 Sb	Te	I	Xe											
6 Cs	Ba	L	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	82 Pb	83 Bi	84 Po	85 At	Rn											
7 Fr	Ra	A											L	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	69 Tm	70 Yb	71 Lu
													A	89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Bring order to this mess ...

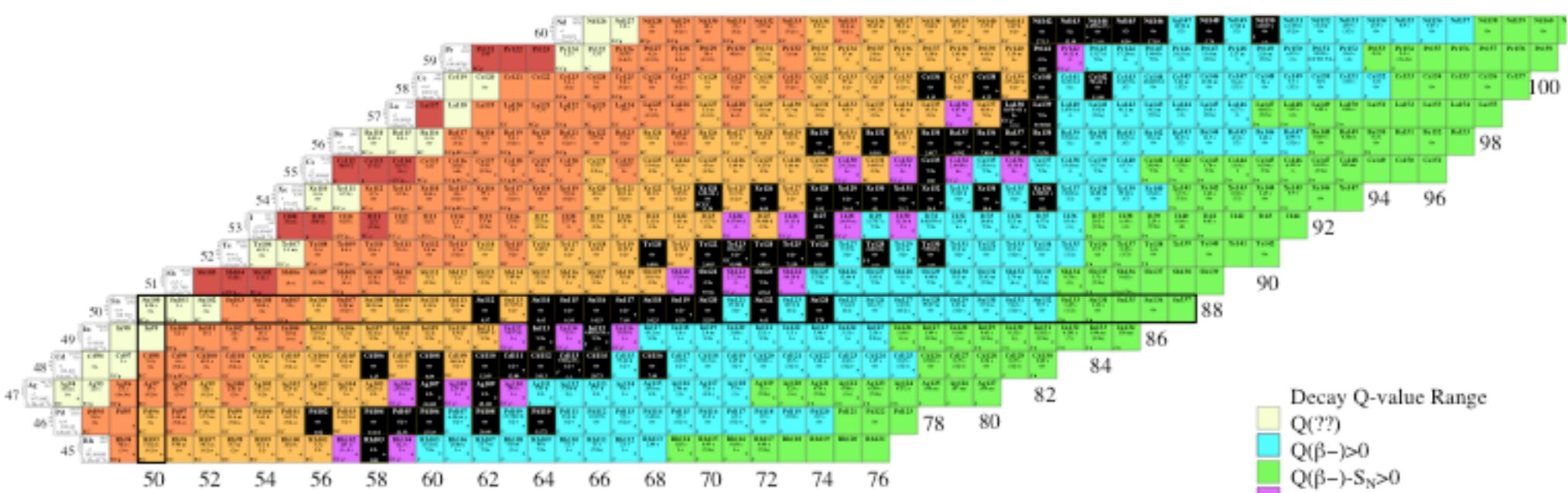
Table of Isotopes



Z=28-45 Part 1 of 3

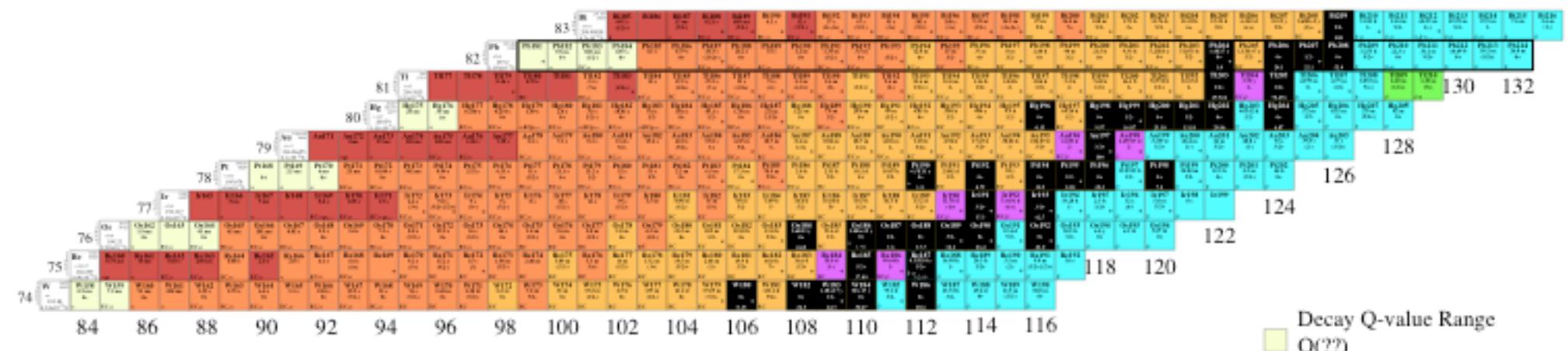


Z=45-60 Part 1 of 3

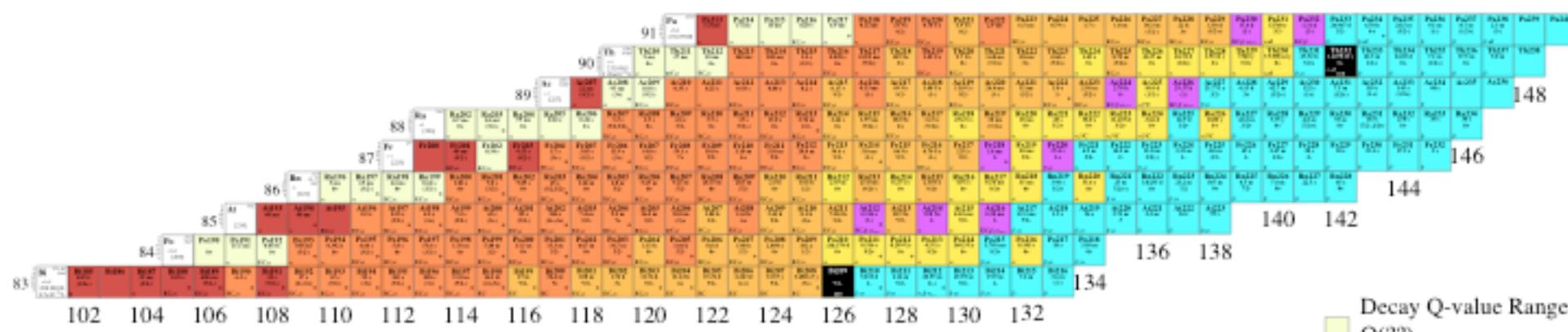


Bring order to this mess ...

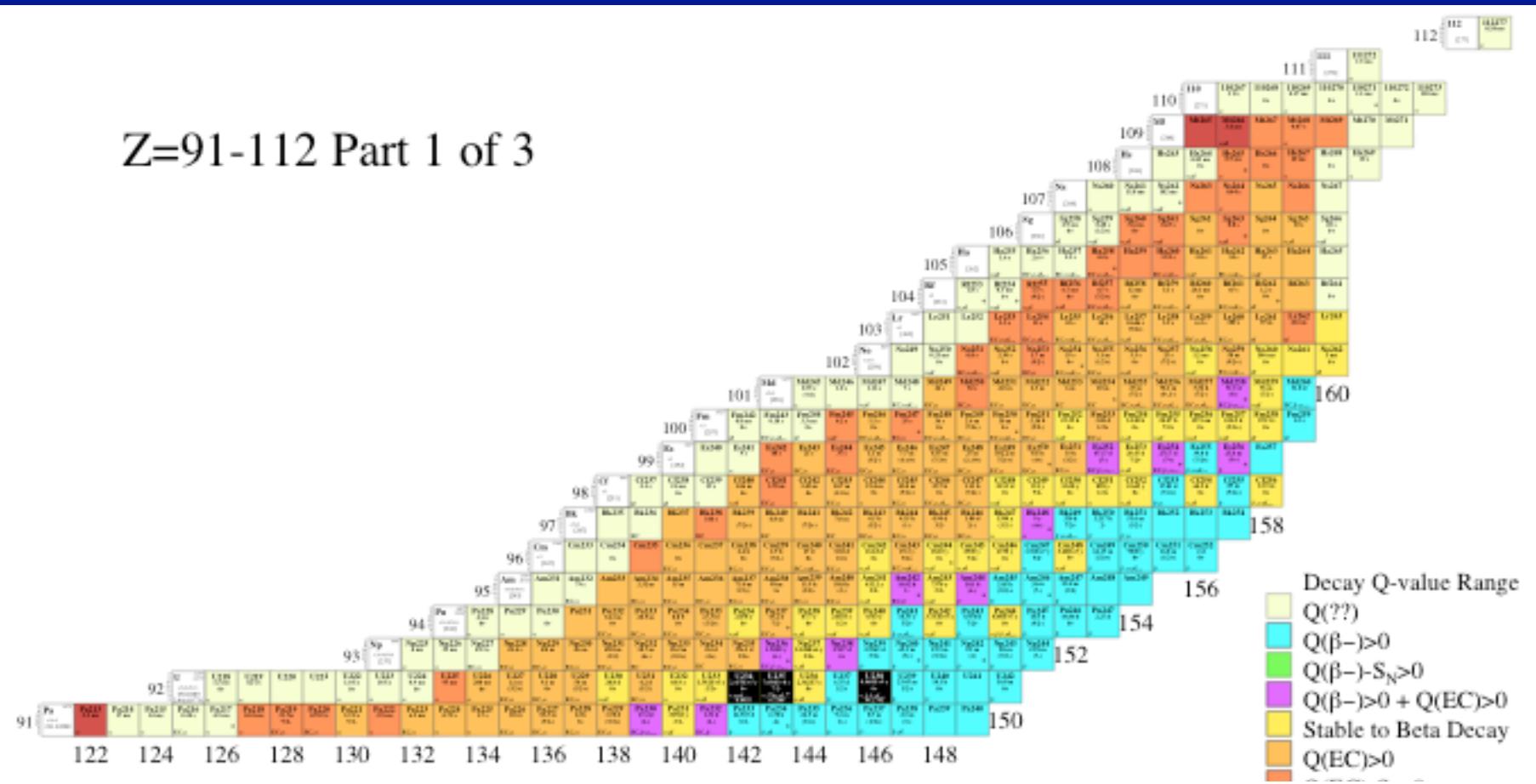
Z=60-74 Part 1 of 4



Z=83-91 Part 1 of 3



Z=91-112 Part 1 of 3



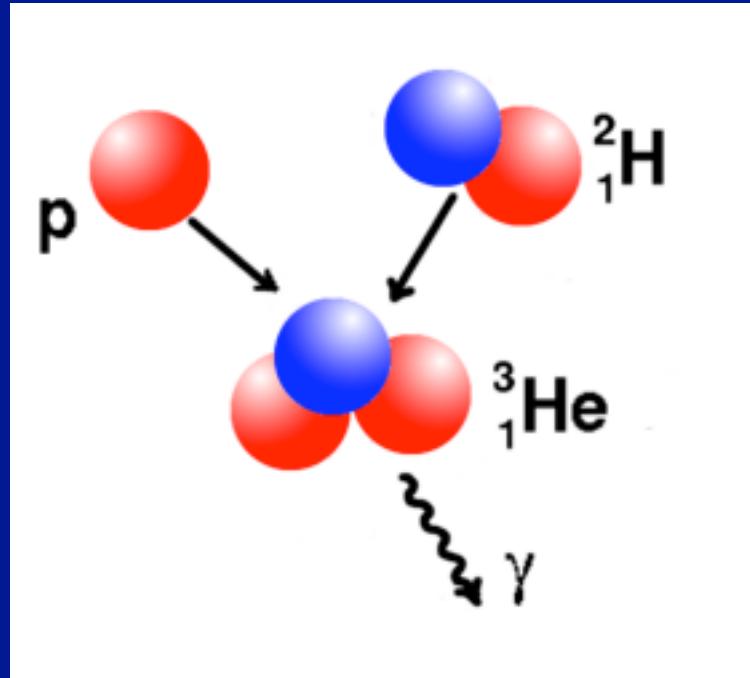
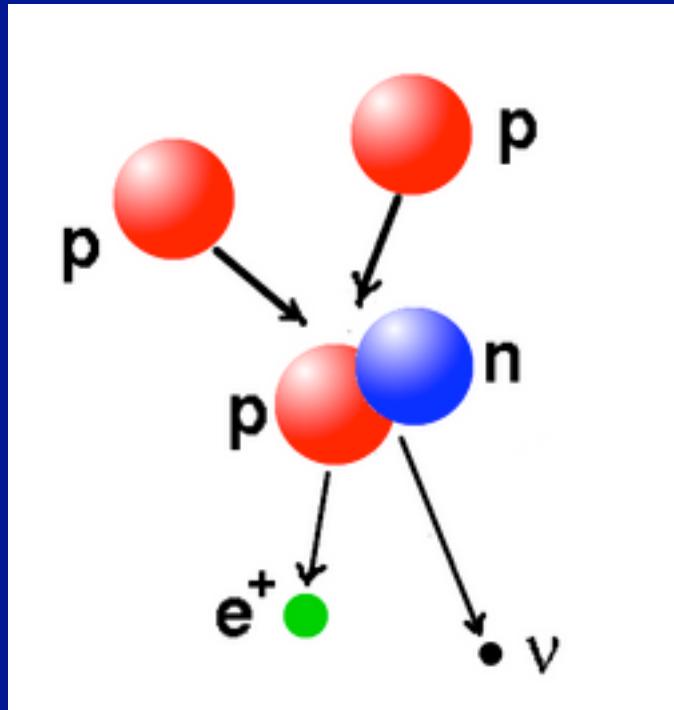
Elements are made by nucleosynthesis in stars'

Nucleosynthesis: Creating new nuclei by nuclear fusion or fission

For example:



High
temperature
required!!!



*OK, not exactly true: H and the vast majority of He (both ${}^3\text{He}$ and ${}^4\text{He}$) and Li is made in the Big Bang

Elements are made by nucleosynthesis in stars

And so on ...:



.....

Title: Evolution and Yields of Extremely Metal Poor Intermediate Mass Stars

Author: Herwig F.

Table: Reaction network

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Byte-by-byte Description of file: datafile1.txt

Bytes	Format	Units	Label	Explanations
1- 37	A37	---	Type	Type
39- 41	I3	---	ID	Identification number
43- 79	A37	---	React	Reaction description
81	A1	---	rReact	Reference for React (1)

Note (1):

- 1 = NACRE adopted;
- 2 = Horiguchi et al. (1996);
- 3 = Caughlan & Fowler (1988);
- 4 = Iliadis et al. (2001);
- 5 = Bao et al. (2000a);
- 6 = Beer et al. (2001);
- 7 = Bao et al. (2000b);
- 8 = Hauser-Feshbach (Jorissen & Goriely 2001);
- 9 = see Jorissen & Goriely (2001);
- a = Wiescher et al. (1990);
- b = Koehler & O'Brien (1989);
- c = Takahashi Yokoi (1987);
- d = Goriely (1999);
- e = Hauser-Feshbach (Jorissen & Goriely 2001).

p-capture reactions	1	2	PROT	(0 00000, 0 00000)	1	DEUT	1
p-capture reactions	2	1	DEUT	(1 PROT , 0 00000)	1	HE 3	1
p-capture reactions	6	1	LI 7	(1 PROT , 0 00000)	2	HE 4	1
p-capture reactions	7	1	BE 7	(1 PROT , 0 00000)	1	B 8	1
p-capture reactions	10	1	C 12	(1 PROT , 0 00000)	1	N 13	1
p-capture reactions	11	1	C 13	(1 PROT , 0 00000)	1	N 14	1
p-capture reactions	12	1	N 14	(1 PROT , 0 00000)	1	O 15	1
p-capture reactions	13	1	N 15	(1 PROT , 1 HE 4)	1	C 12	1
p-capture reactions	14	1	N 15	(1 PROT , 0 00000)	1	O 16	1
p-capture reactions	15	1	O 16	(1 PROT , 0 00000)	1	F 17	1
p-capture reactions	16	1	O 17	(1 PROT , 1 HE 4)	1	N 14	1
p-capture reactions	17	1	O 17	(1 PROT , 0 00000)	1	F 18	1
p-capture reactions	18	1	O 18	(1 PROT , 1 HE 4)	1	N 15	1
p-capture reactions	19	1	O 18	(1 PROT , 0 00000)	1	F 19	1
p-capture reactions	20	1	F 19	(1 PROT , 1 HE 4)	1	O 16	1
p-capture reactions	21	1	F 19	(1 PROT , 0 00000)	1	NE 20	1
p-capture reactions	22	1	NE 20	(1 PROT , 0 00000)	1	NA 21	1
p-capture reactions	23	1	NE 21	(1 PROT , 0 00000)	1	NA 22	1
p-capture reactions	26	1	NA 22	(1 PROT , 0 00000)	1	MG 23	1
p-capture reactions	30	1	MG 25	(1 PROT , 0 00000)	1	AL 26g	1

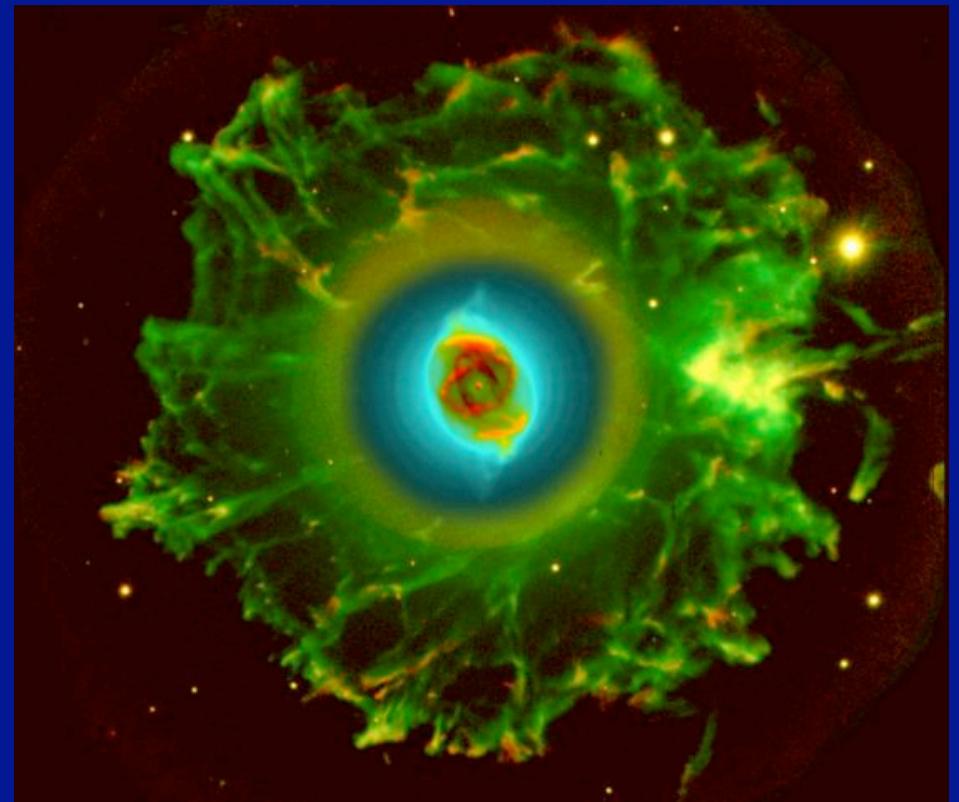
Elements are made by nucleosynthesis in stars

p-capture reactions	33 1 AL 26g(1 PROT , 0 OOOOO)	1 SI 27 4	n-capture reactions	58 1 C 12 (1 NEUT , 0 OOOOO)
p-capture reactions	35 1 AL 27 (1 PROT , 1 HE 4)	1 MG 24 1	n-capture reactions	62 1 NE 21 (1 NEUT , 0 OOOOO)
p-capture reactions	37 1 SI 28 (1 PROT , 0 OOOOO)	1 P 29 1	n-capture reactions	64 1 NE 22 (1 NEUT , 0 OOOOO)
p-capture reactions	38 1 SI 29 (1 PROT , 0 OOOOO)	1 P 30 1	n-capture reactions	85 1 N 14 (1 NEUT , 1 PROT)
p-capture reactions	39 1 SI 30 (1 PROT , 0 OOOOO)	1 P 31 1	n-capture reactions	86 1 SI 28 (1 NEUT , 0 OOOOO)
p-capture reactions	63 1 NE 22 (1 PROT , 0 OOOOO)	1 NA 23 1	n-capture reactions	87 1 SI 29 (1 NEUT , 0 OOOOO)
p-capture reactions	65 1 NA 23 (1 PROT , 0 OOOOO)	1 MG 24 1	n-capture reactions	88 1 SI 30 (1 NEUT , 0 OOOOO)
p-capture reactions	66 1 NA 23 (1 PROT , 1 HE 4)	1 NE 20 1	n-capture reactions	89 1 N 14 (1 NEUT , 0 OOOOO)
p-capture reactions	69 1 MG 24 (1 PROT , 0 OOOOO)	1 AL 25 1	n-capture reactions	90 1 O 16 (1 NEUT , 0 OOOOO)
p-capture reactions	70 1 MG 25 (1 PROT , 0 OOOOO)	1 AL 26 1	n-capture reactions	93 1 FE 56 (1 NEUT , 0 OOOOO)
p-capture reactions	72 1 MG 26 (1 PROT , 0 OOOOO)	1 AL 27 1	n-capture reactions	94 1 FE 57 (1 NEUT , 0 OOOOO)
p-capture reactions	73 1 AL 27 (1 PROT , 0 OOOOO)	1 SI 28 1	n-capture reactions	95 1 FE 58 (1 NEUT , 0 OOOOO)
p-capture reactions	74 1 AL 27 (1 PROT , 1 HE 4)	1 MG 24 1	n-capture reactions	96 1 CO 59 (1 NEUT , 0 OOOOO)
p-capture reactions	79 1 B 11 (1 PROT , 0 OOOOO)	3 HE 4 1	n-capture reactions	97 1 NI 60 (1 NEUT , 0 OOOOO)
p-capture reactions	81 1 C 14 (1 PROT , 0 OOOOO)	1 N 15 a	n-capture reactions	98 1 NI 61 (1 NEUT , 0 OOOOO)
p-capture reactions	92 1 N 13 (1 PROT , 0 OOOOO)	1 O 14 1	n-capture reactions	100 1 P 31 (1 NEUT , 0 OOOOO)
p-capture reactions	101 1 P 31 (1 PROT , 0 OOOOO)	1 S 32 4	n-capture reactions	106 1 NI 58 (1 NEUT , 0 OOOOO)
{alpha} - and ^3^He-capture reactions	3 2 HE 3 (0 OOOOO, 2 PROT)	1 HE 4 1	n-capture reactions	107 1 NI 59 (1 NEUT , 0 OOOOO)
{alpha} - and ^3^He-capture reactions	4 1 HE 4 (1 HE 3, 0 OOOOO)	1 BE 7 1	n-capture reactions	108 1 NI 59 (1 NEUT , 1 PROT)
{alpha} - and ^3^He-capture reactions	40 3 HE 4 (0 OOOOO, 0 OOOOO)	1 C 12 1	{beta} -decay and e^-^-captures	5 1 BE 7 (0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	41 1 C 12 (1 HE 4, 0 OOOOO)	1 O 16 1	{beta} -decay and e^-^-captures	8 1 B 8 (0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	42 1 C 13 (1 HE 4, 1 NEUT)	1 O 16 1	{beta} -decay and e^-^-captures	9 1 B 8 (0 OOOOO, 1 PROT)
{alpha} - and ^3^He-capture reactions	43 1 N 14 (1 HE 4, 0 OOOOO)	1 F 18 1	{beta} -decay and e^-^-captures	25 1 NA 22 (0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	44 1 O 16 (1 HE 4, 0 OOOOO)	1 NE 20 1	{beta} -decay and e^-^-captures	34 1 AL 26g(0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	45 1 O 18 (1 HE 4, 0 OOOOO)	1 NE 22 1	{beta} -decay and e^-^-captures	77 1 C 11 (0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	46 1 NE 20 (1 HE 4, 0 OOOOO)	1 MG 24 1	{beta} -decay and e^-^-captures	91 1 N 13 (0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	48 1 NE 21 (1 HE 4, 1 NEUT)	1 MG 24 1	{beta} -decay and e^-^-captures	104 1 C 14 (0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	49 1 NE 22 (1 HE 4, 0 OOOOO)	1 MG 26 1	{beta} -decay and e^-^-captures	105 1 NI 59 (0 OOOOO, 0 OOOOO)
{alpha} - and ^3^He-capture reactions	50 1 NE 22 (1 HE 4, 1 NEUT)	1 MG 25 1	C-burning	59 2 C 12 (0 OOOOO, 1 PROT)
{alpha} - and ^3^He-capture reactions	53 1 MG 24 (1 HE 4, 0 OOOOO)	1 SI 28 3	C-burning	60 2 C 12 (0 OOOOO, 1 HE)
{alpha} - and ^3^He-capture reactions	61 1 NE 20 (1 HE 4, 1 PROT)	1 NA 23 1		
{alpha} - and ^3^He-capture reactions	67 1 NA 23 (1 HE 4, 1 PROT)	1 MG 26 8		
{alpha} - and ^3^He-capture reactions	68 1 MG 24 (1 HE 4, 1 PROT)	1 AL 27 1		
{alpha} - and ^3^He-capture reactions	71 1 MG 25 (1 HE 4, 1 NEUT)	1 SI 28 1		
{alpha} - and ^3^He-capture reactions	75 1 BE 7 (1 HE 4, 0 OOOOO)	1 C 11 1		
{alpha} - and ^3^He-capture reactions	76 1 LI 7 (1 HE 4, 0 OOOOO)	1 B 11 1		
{alpha} - and ^3^He-capture reactions	78 1 B 11 (1 HE 4, 1 NEUT)	1 N 14 3		
{alpha} - and ^3^He-capture reactions	80 1 C 14 (1 HE 4, 0 OOOOO)	1 O 18 9		
{alpha} - and ^3^He-capture reactions	82 1 N 15 (1 HE 4, 0 OOOOO)	1 F 19 1		
{alpha} - and ^3^He-capture reactions	83 1 O 17 (1 HE 4, 0 OOOOO)	1 NE 21 3		
{alpha} - and ^3^He-capture reactions	84 1 O 17 (1 HE 4, 1 NEUT)	1 NE 20 1		
n-capture reactions	31 1 AL 26g(1 NEUT , 1 PROT)	1 MG 26 3		
n-capture reactions	32 1 AL 26g(1 NEUT , 1 HE 4)	1 NA 23 3		
n-capture reactions	36 1 BE 7 (1 NEUT , 1 PROT)	1 LI 7 0		
n-capture reactions	47 1 NE 20 (1 NEUT , 0 OOOOO)	1 NE 21 5		
n-capture reactions	51 1 NE 22 (1 NEUT , 0 OOOOO)	1 NA 23 6		
n-capture reactions	52 1 NA 23 (1 NEUT , 0 OOOOO)	1 MG 24 5		
n-capture reactions	54 1 MG 24 (1 NEUT , 0 OOOOO)	1 MG 25 5		
n-capture reactions	55 1 MG 25 (1 NEUT , 0 OOOOO)	1 MG 26 5		
n-capture reactions	56 1 MG 26 (1 NEUT , 0 OOOOO)	1 AL 27 5		
n-capture reactions	57 1 AL 27 (1 NEUT , 0 OOOOO)	1 SI 28 5		

How do we know the abundances of elements in stars



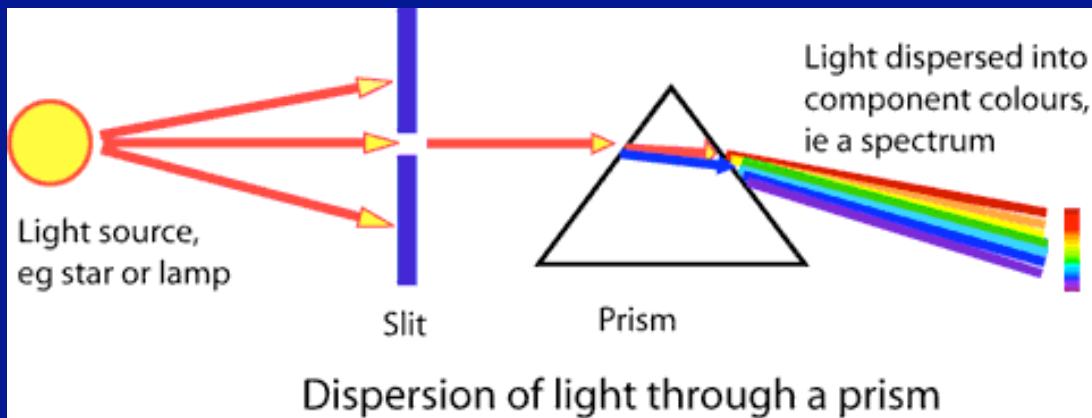
Little Ghost Nebula (NGC 6369), ionized oxygen, hydrogen, and nitrogen atoms are colored blue, green, and red



Cat's Eye Nebula (NGC 6543), emission from nitrogen atoms as red and oxygen atoms as green and blue shades.

How do we know the abundances of elements in stars

1. Stellar spectroscopy:



Types of spectra:

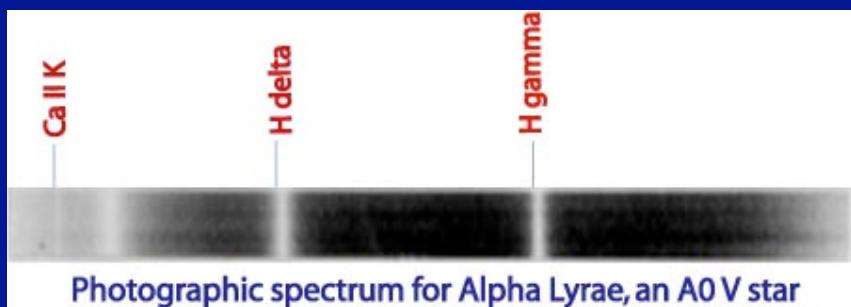
Continuous (or continuum):



Absorption (dark line):

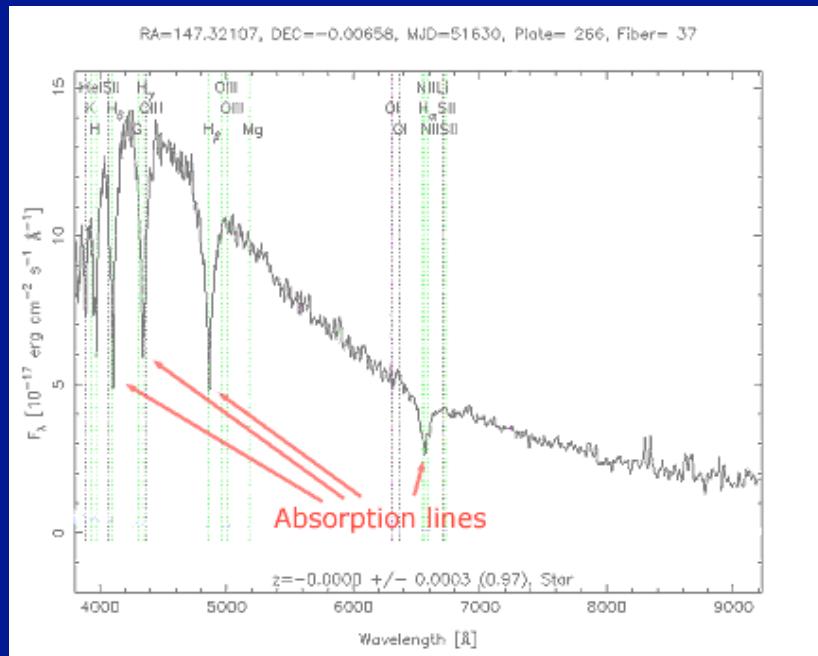


Emission (bright line):

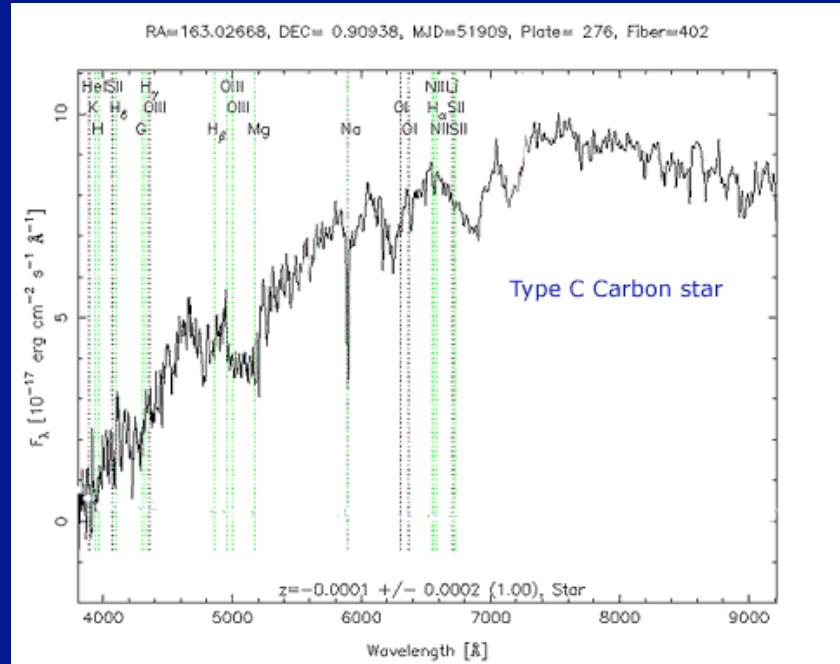


How do we know the abundances of elements in stars

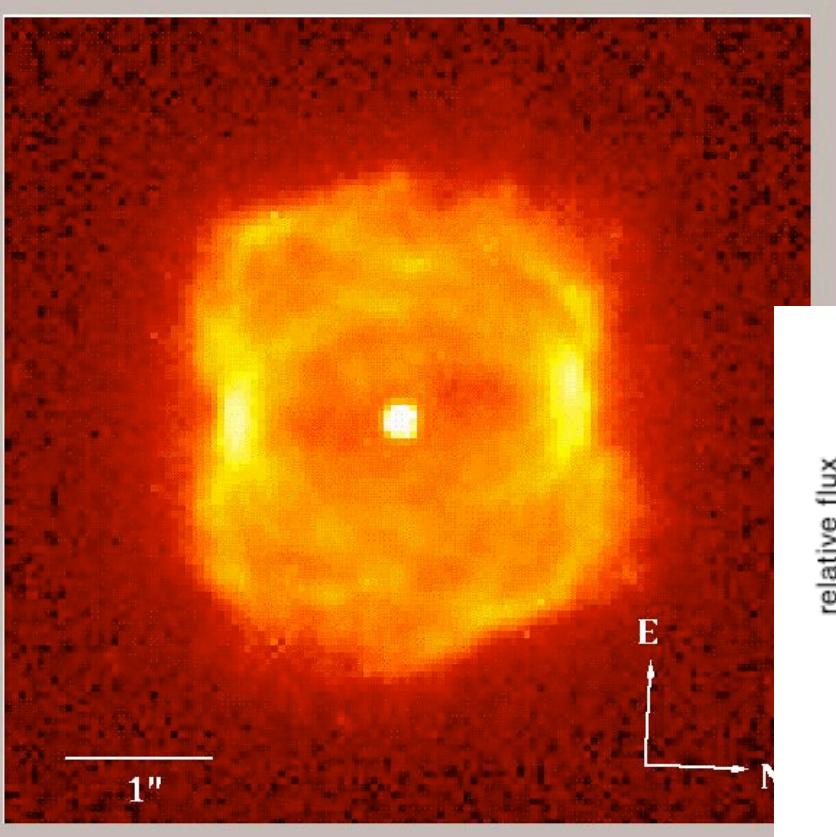
A-type main sequence
Star, hot: H-burning in the core



Giant star: C-rich, cool, inert C/O core, He- and H-shell burning



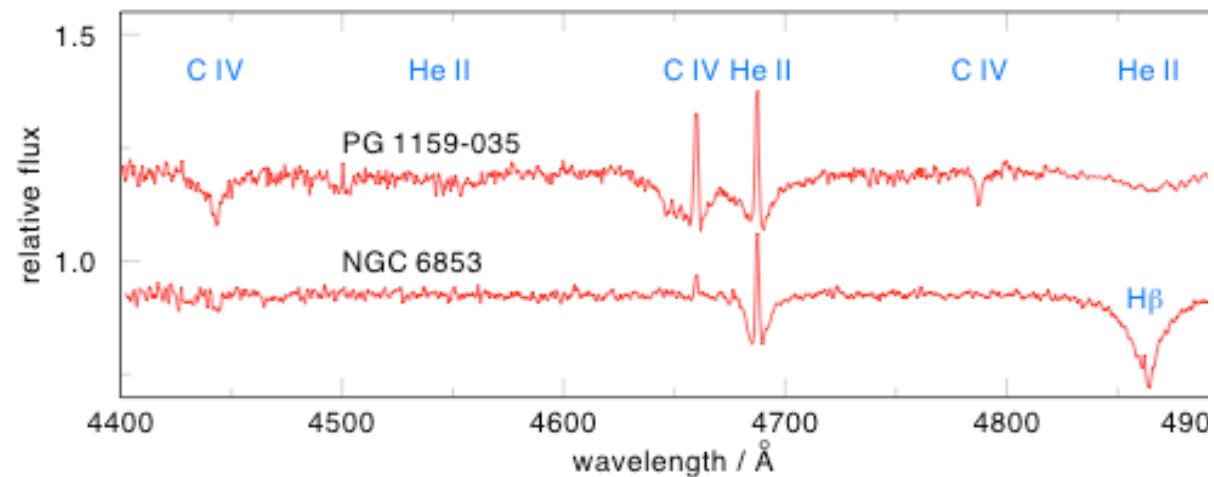
Some exotic stars: hot H-poor central stars of planetary nebulae



PN M4-18

abundances of the central star:

(H,He,C,O) = ($<0.02, 0.46, 0.50, 0.08$)
Sun: ($0.70, 0.28, 0.003, 0.008$)



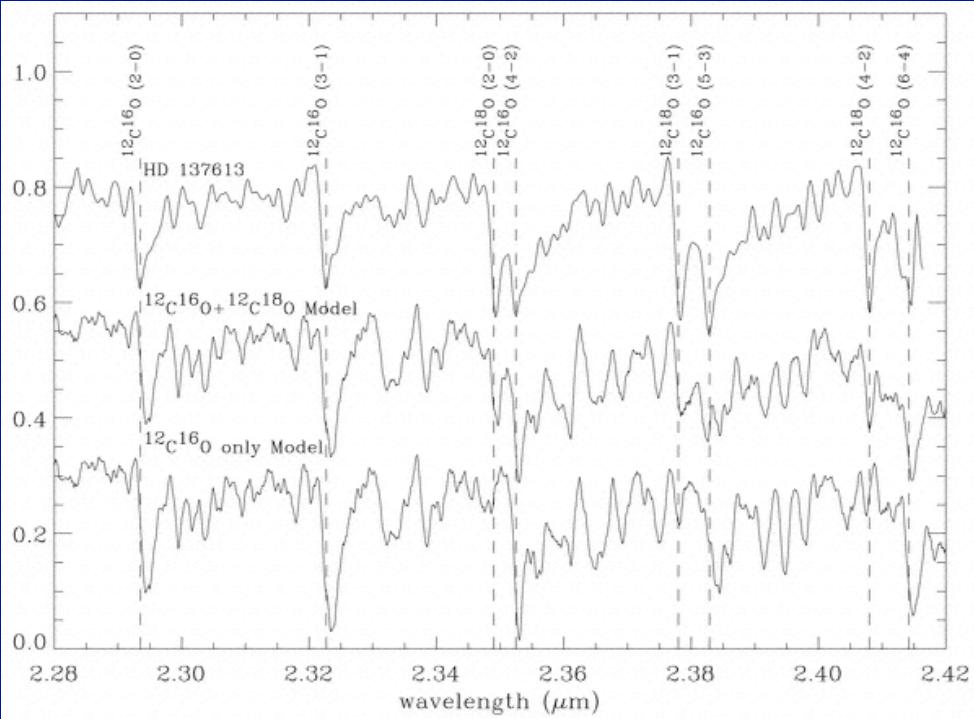
Hubble Space Telescope image Spectroscopy at 3.5m telescope Calar Alto, Sp

Some exotic stars: A H-deficient ^{18}O -rich C-star

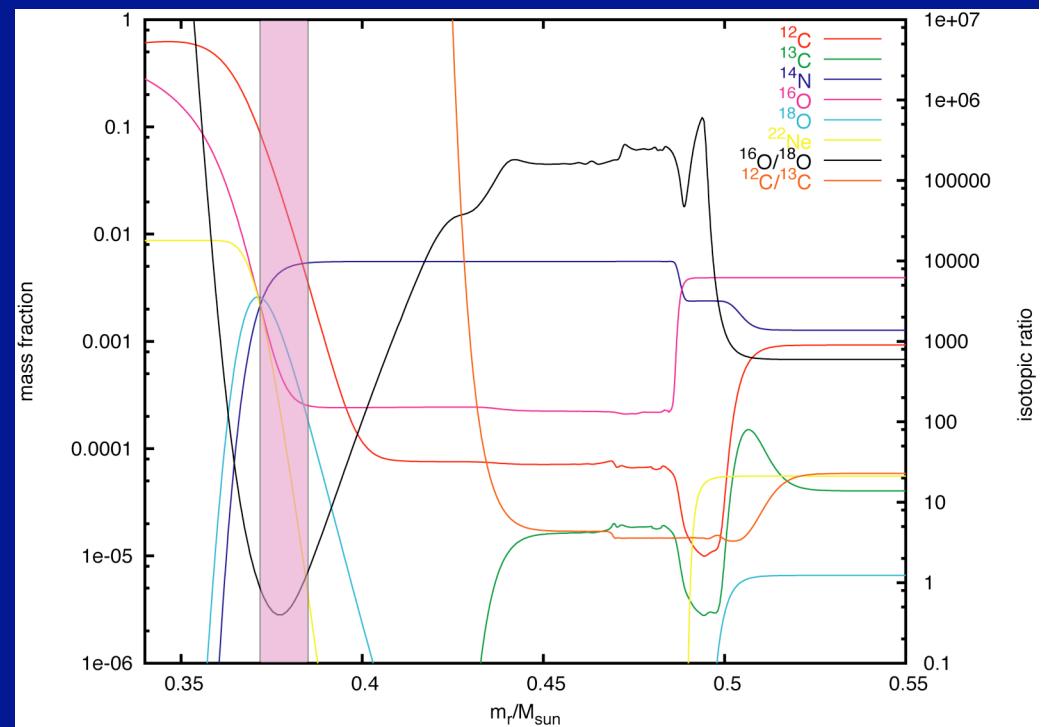
HD 137613
Sun

$^{16}\text{O}/^{18}\text{O}$
 < 1.0
440

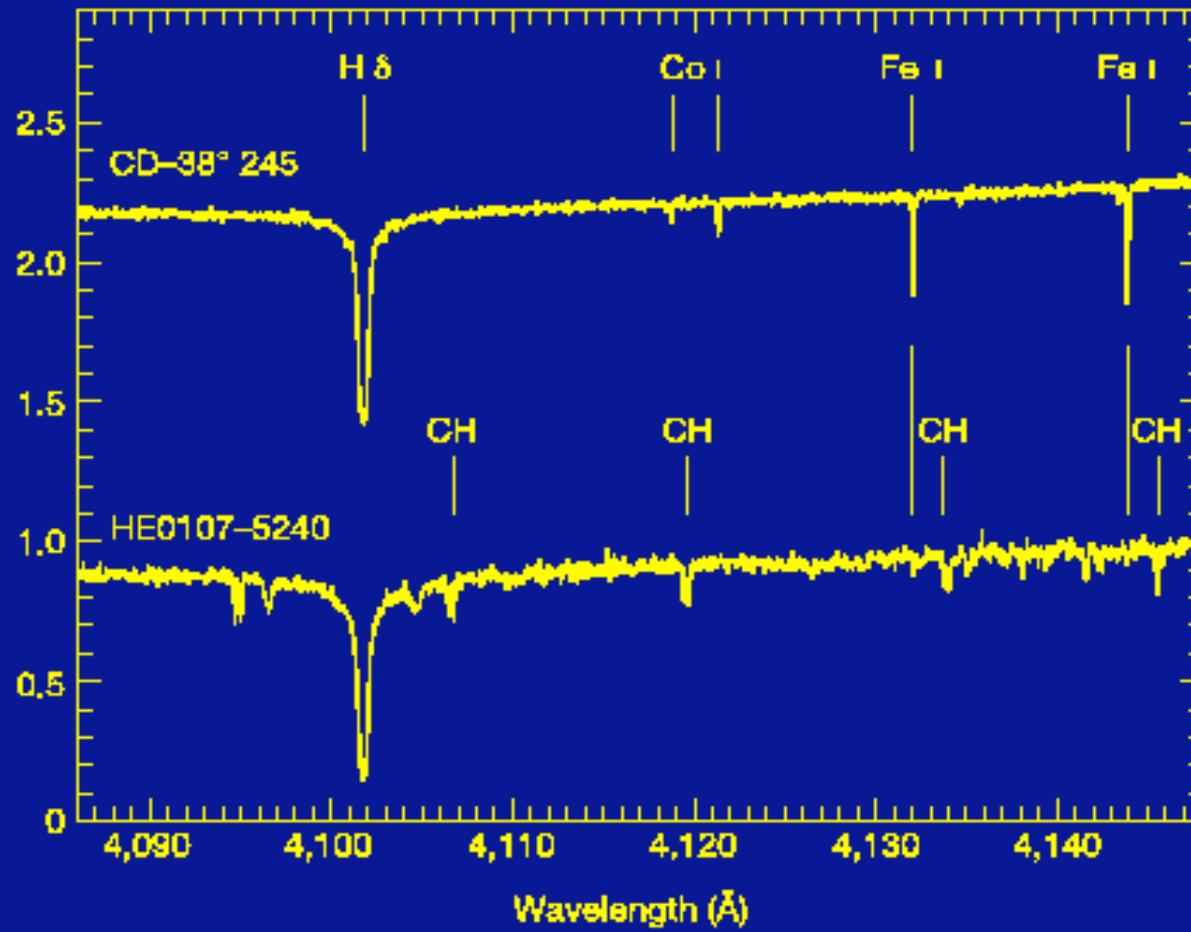
HD 137513



Abundance profile in the interior of a 2M_\odot stellar model after the completion of H-core and He-core burning.



The most metal-poor star: Messenger from the early universe

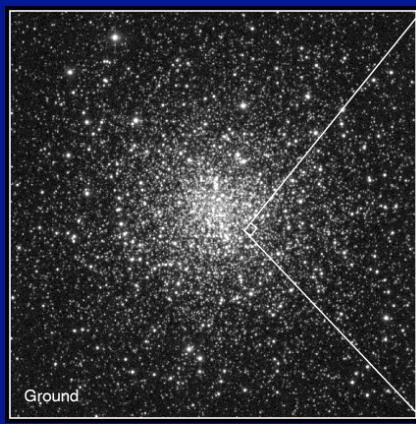


Metal content 100.000 times smaller than in the sun: $[Fe/H] = -5.3$

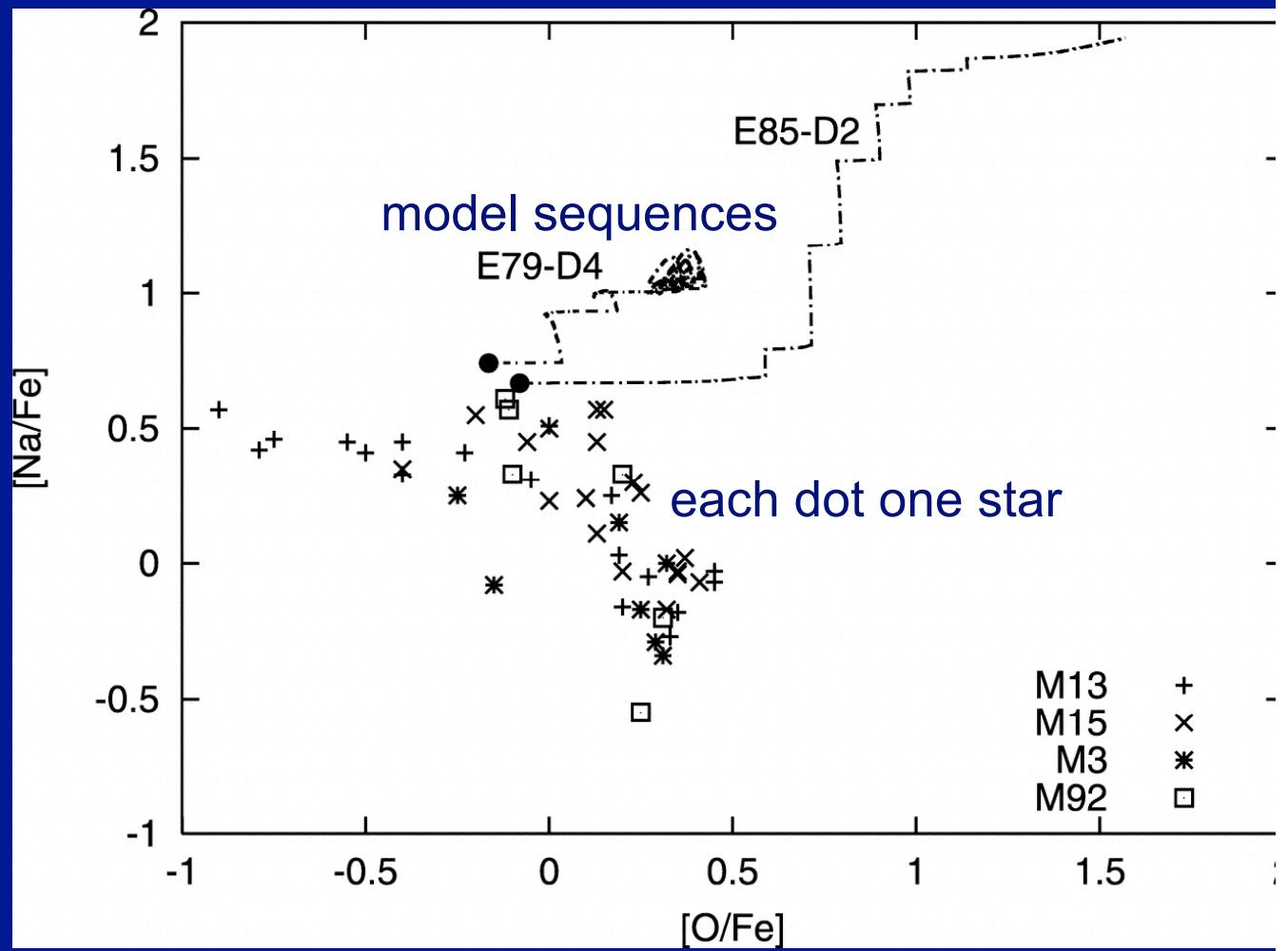
This star has formed shortly after the Big Bang when the Universe was young.

Abundances of many stars in one population:

The O-Na anti-correlation in globular cluster stars

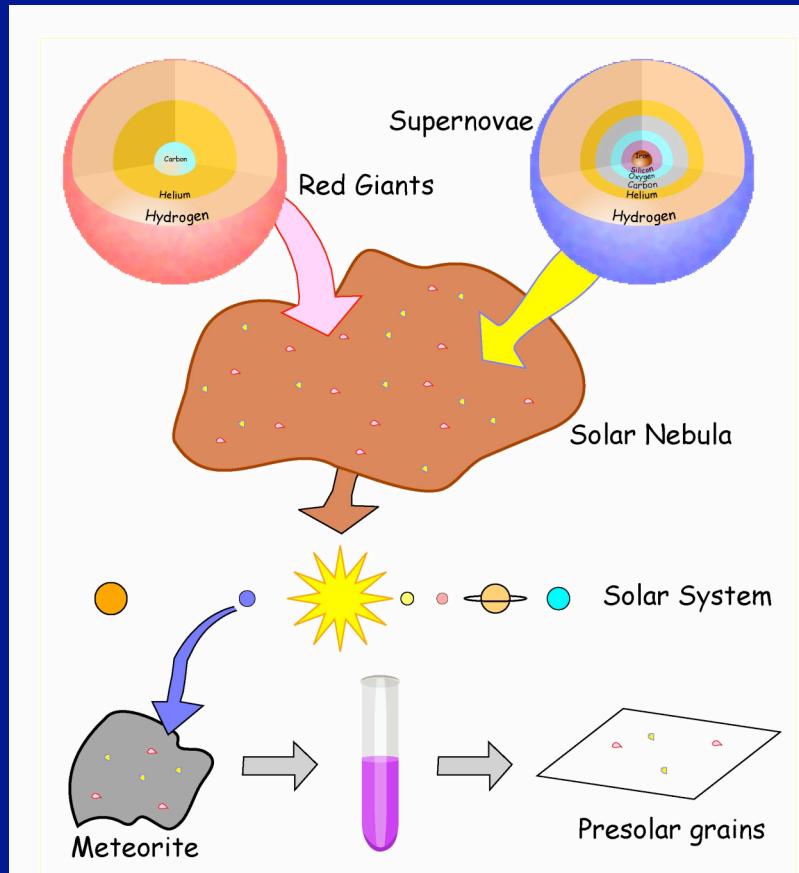


M4

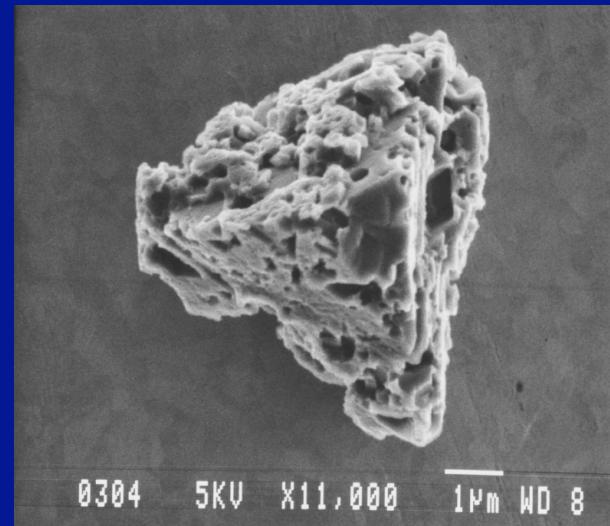


Stellar isotopic abundances from pre-solar grains:

Dust forms in the cool mass-loss outflows of stars and the supernova debris.

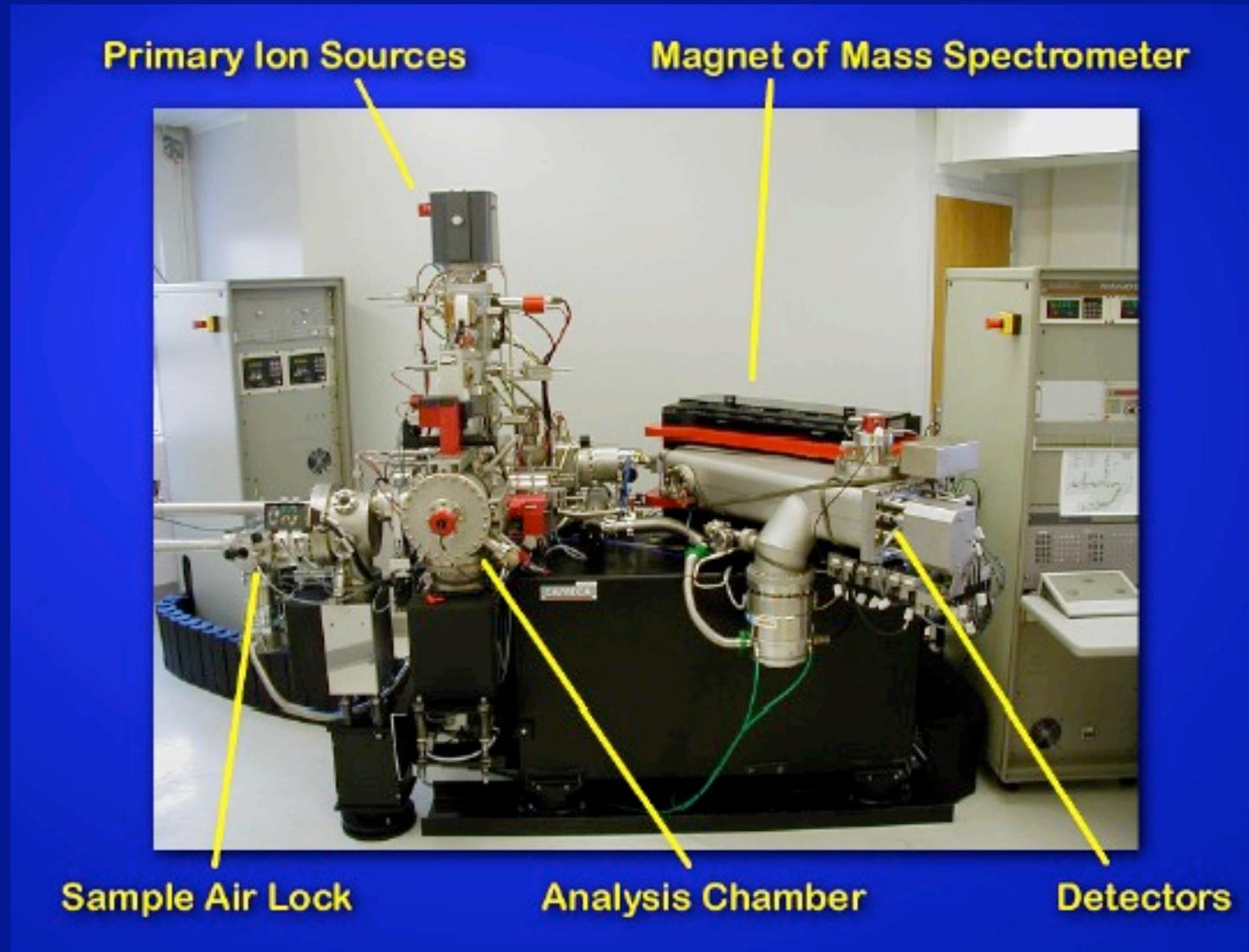


Individual dust grains extracted from primitive meteorites can be associated with their individual site of origin around one star ... tracing that star's individual isotopic signature.



Stellar isotopic abundances from pre-solar grains:

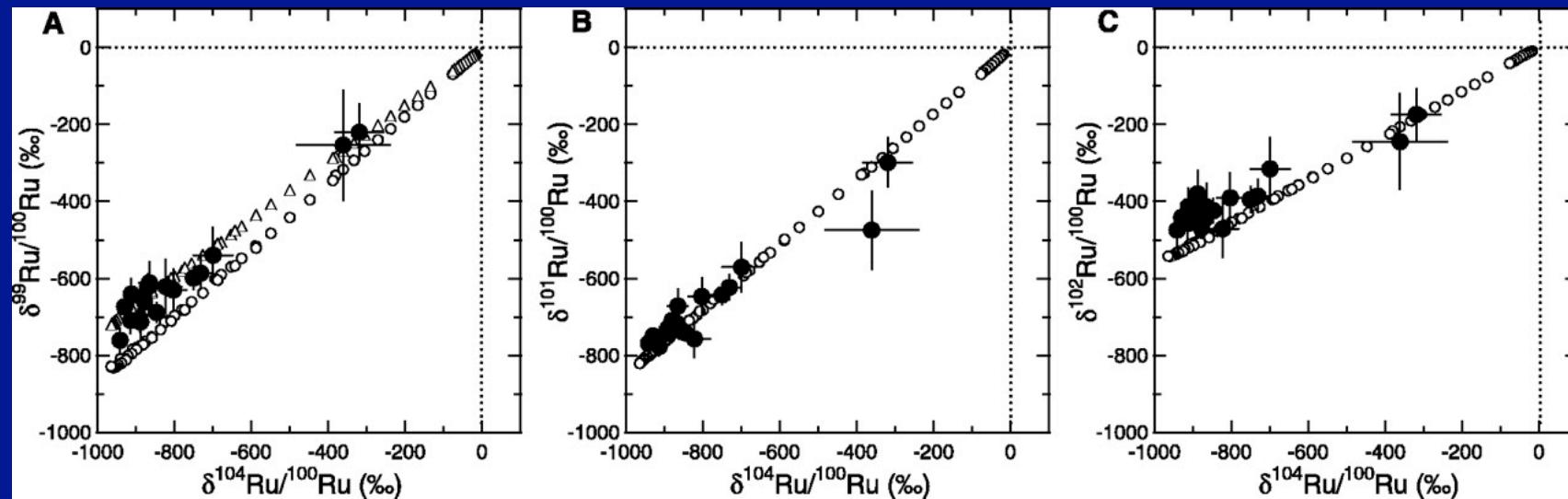
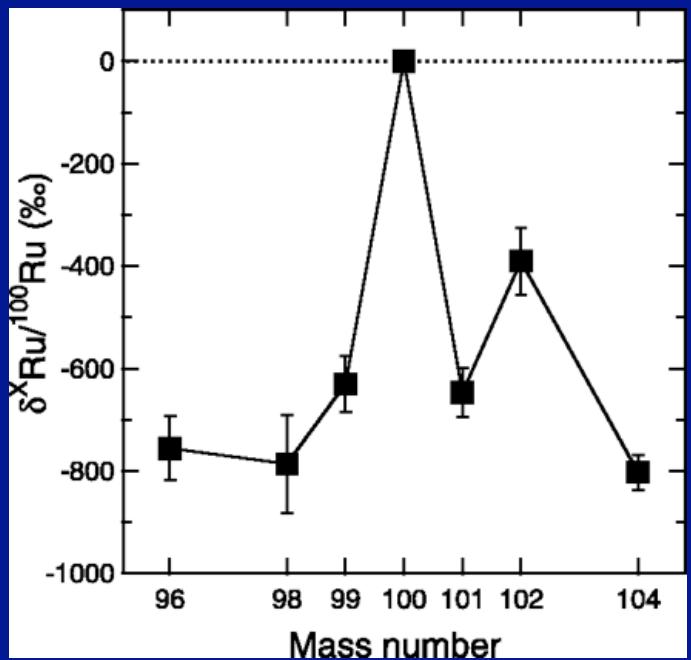
The nano-SIMS: nano-meter secondary ion-mass spectroscopy



One at Washington University, St. Louis, one at Max-Planck Insitut for cosmo-chemistry in Mainz, Germany.

Stellar isotopic abundances from pre-solar grains:

*Extinct Technetium in Silicon Carbide Stardust Grains:
Implications for Stellar Nucleosynthesis*
Savina et al. 2004, Science 303, 649



Outline:

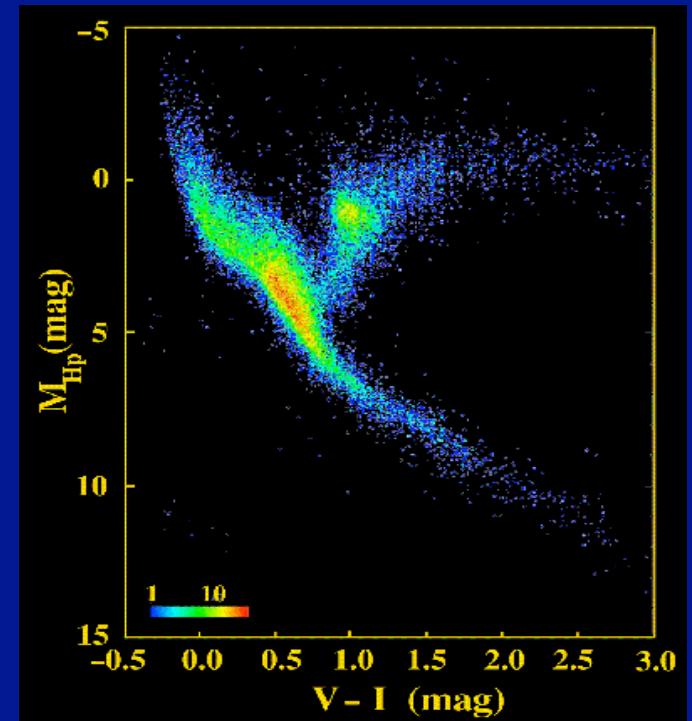
1. What is an element?
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 - o Stellar spectra
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5. Scientists paradise: new data, lots of it!
6. Conclusions

Calculating the evolution of stars and the making of the elements

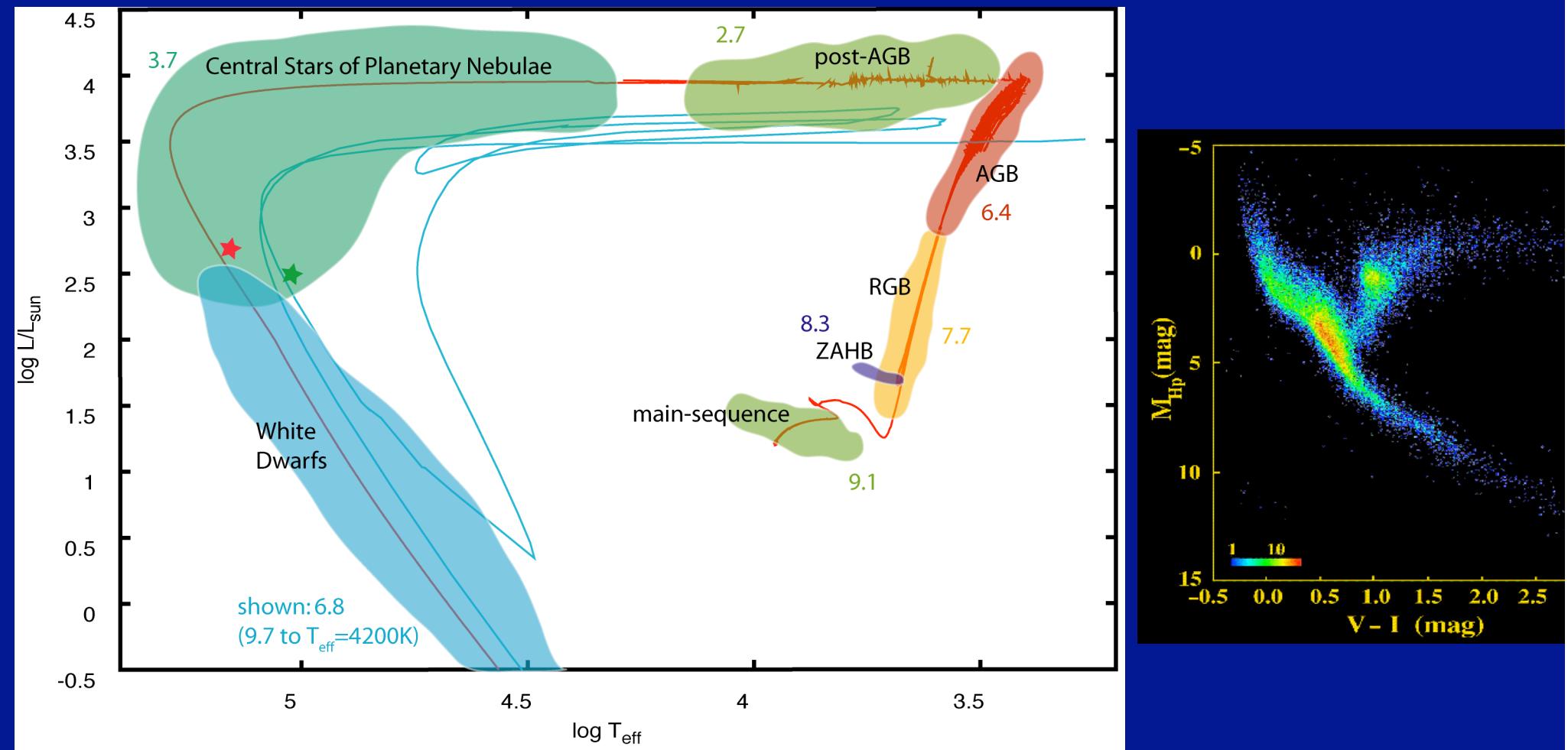
1D evolution of stellar structure

Time sequence of stellar models, each model:

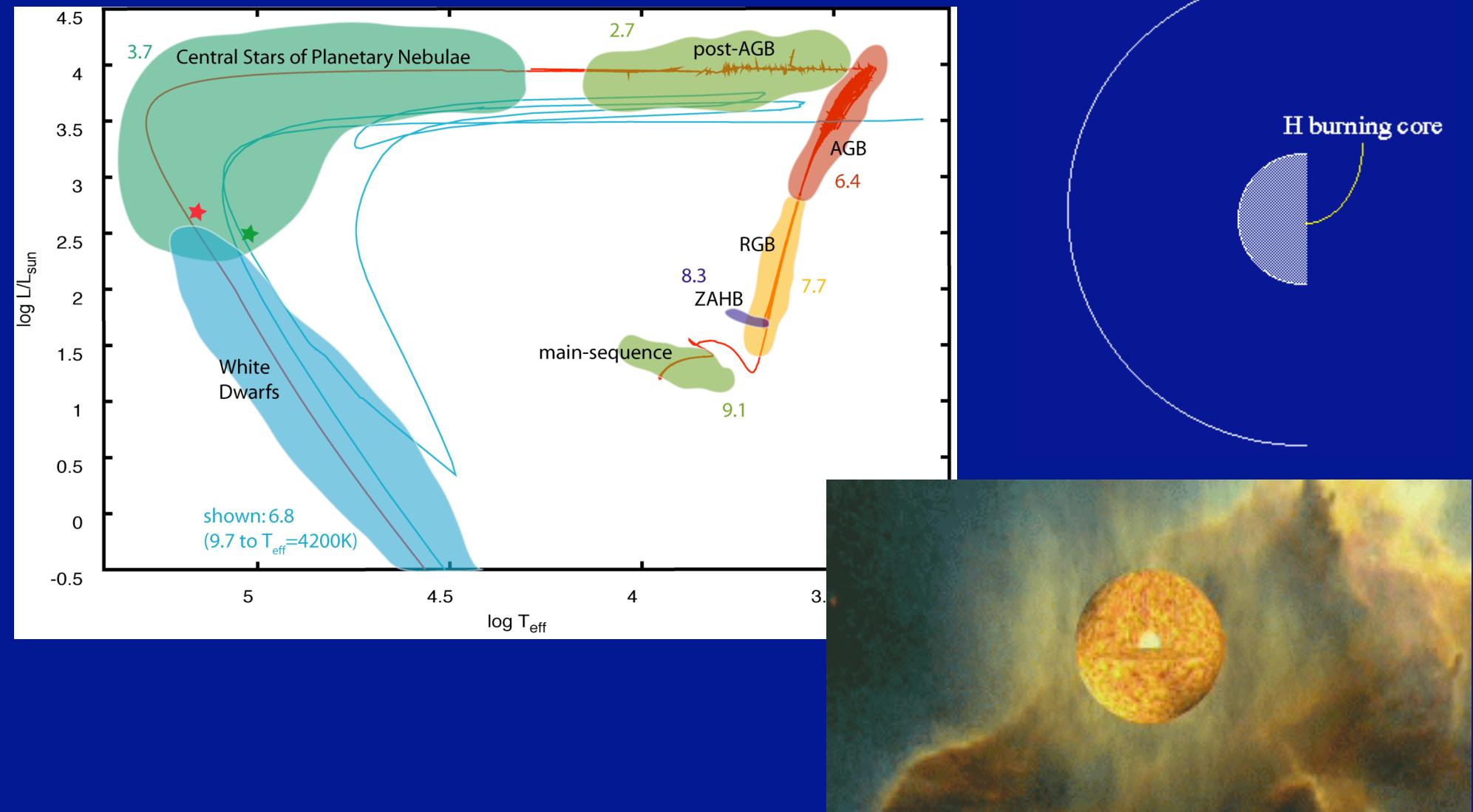
- Spherically symmetric
- Governing equations:
 - mass conservation
 - hydrostatic equilibrium
 - energy equation
 - energy transport
- Material properties
 - opacities
 - equation of state (EOS)
- Mixing, for example due to convection
- Nucleosynthesis
- Evolution through the complete life cycle of stars



The life cycle of stars: Evolving through a succession of stages

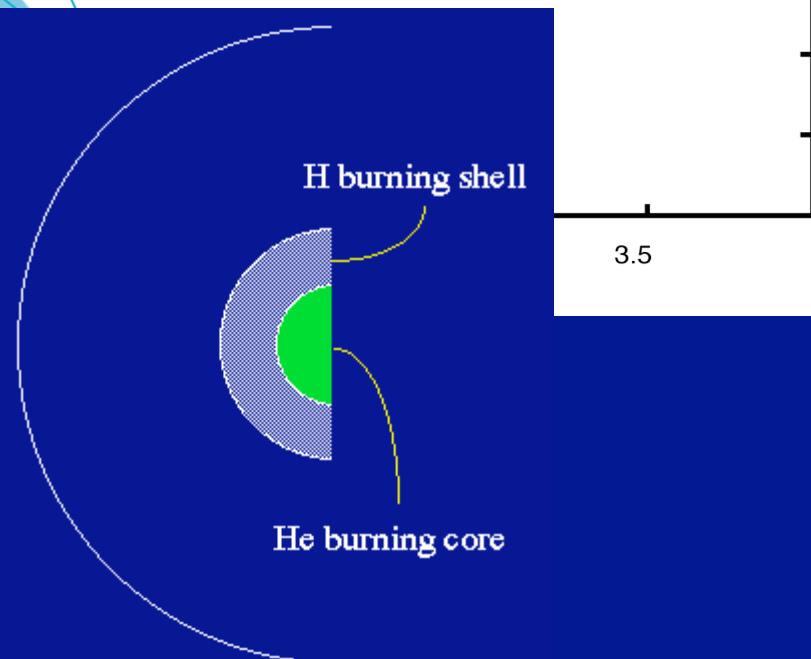
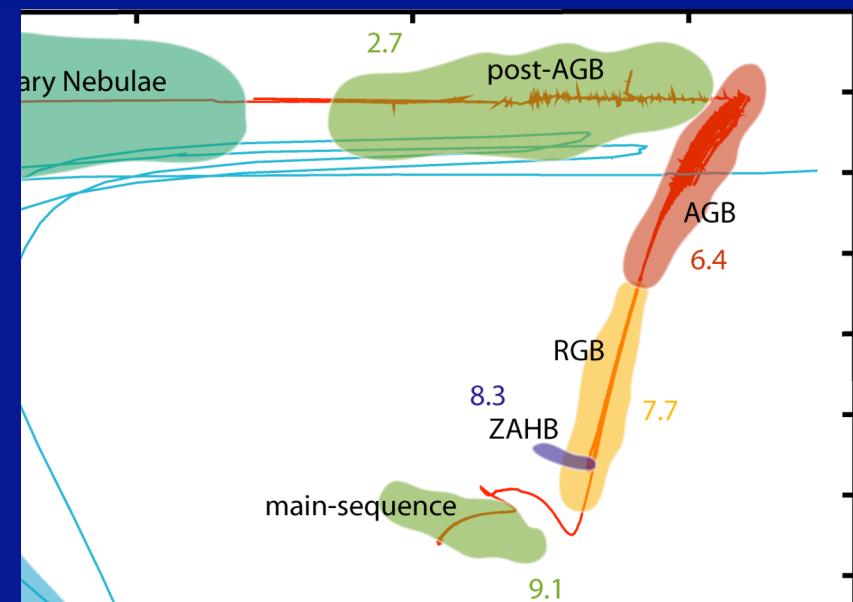
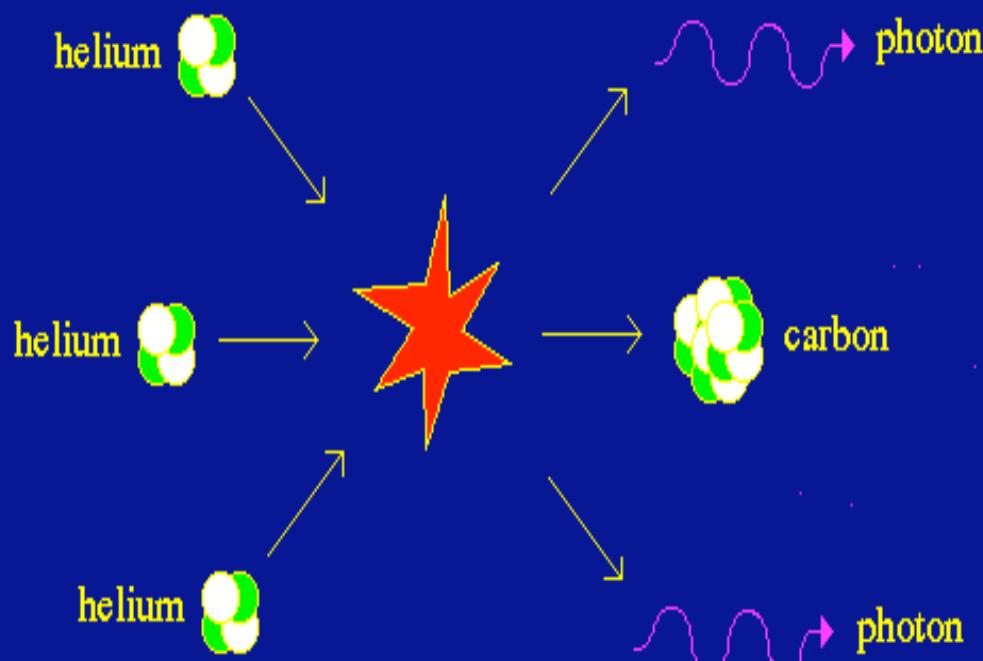


The life cycle of stars: main sequence - H-core burning

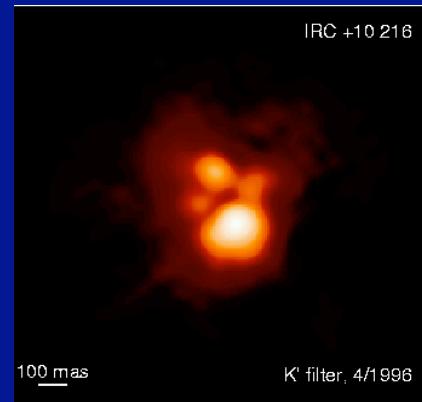


The life cycle of stars:

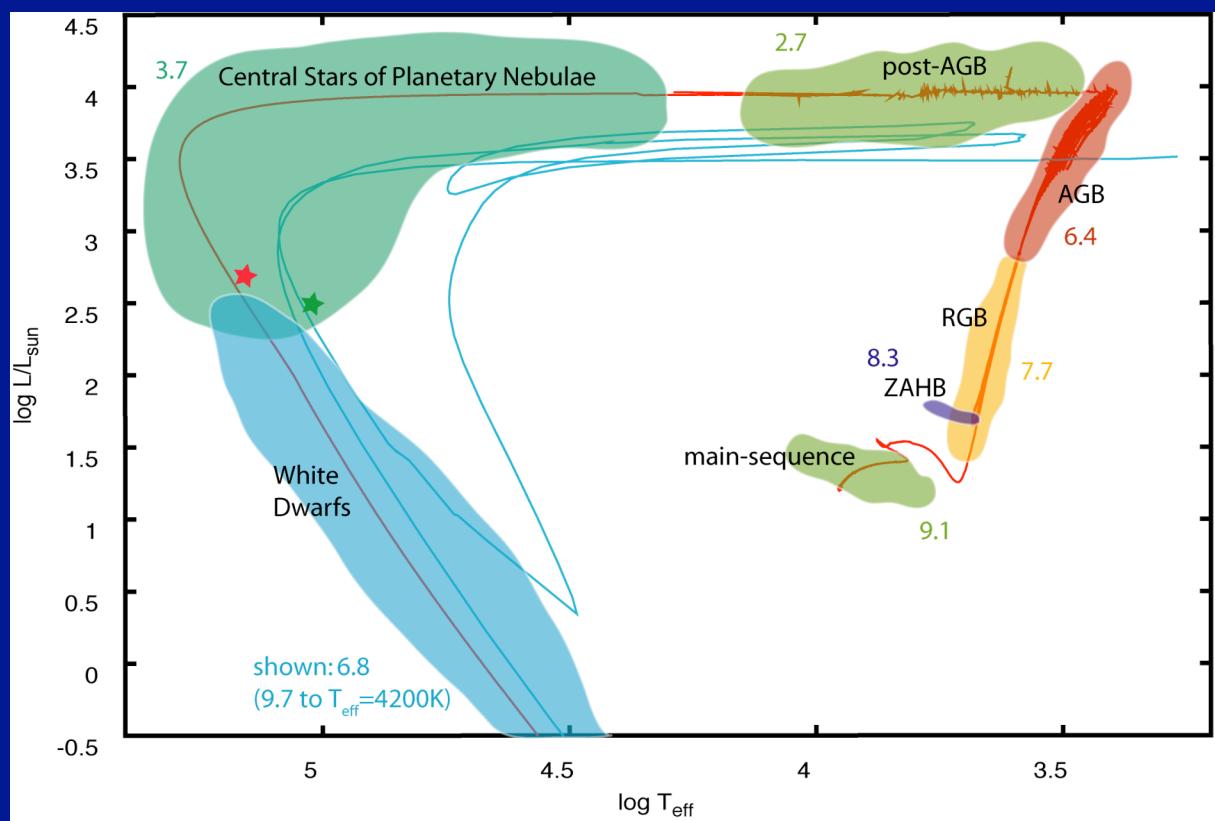
ZAHB - He-core burning



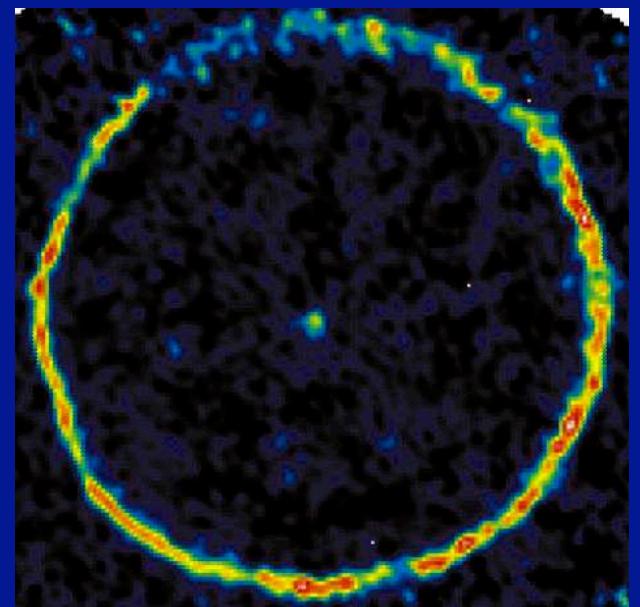
The life cycle of stars: Asymptotic Giant Branch H- and He-shell burning



Interferometric
image of AGB C-
star IRC 10+216



Radio emission of CO around
AGB star TT Cygni



Surface Abundance Evolution of AGB stars with low metal content:

initial mass:

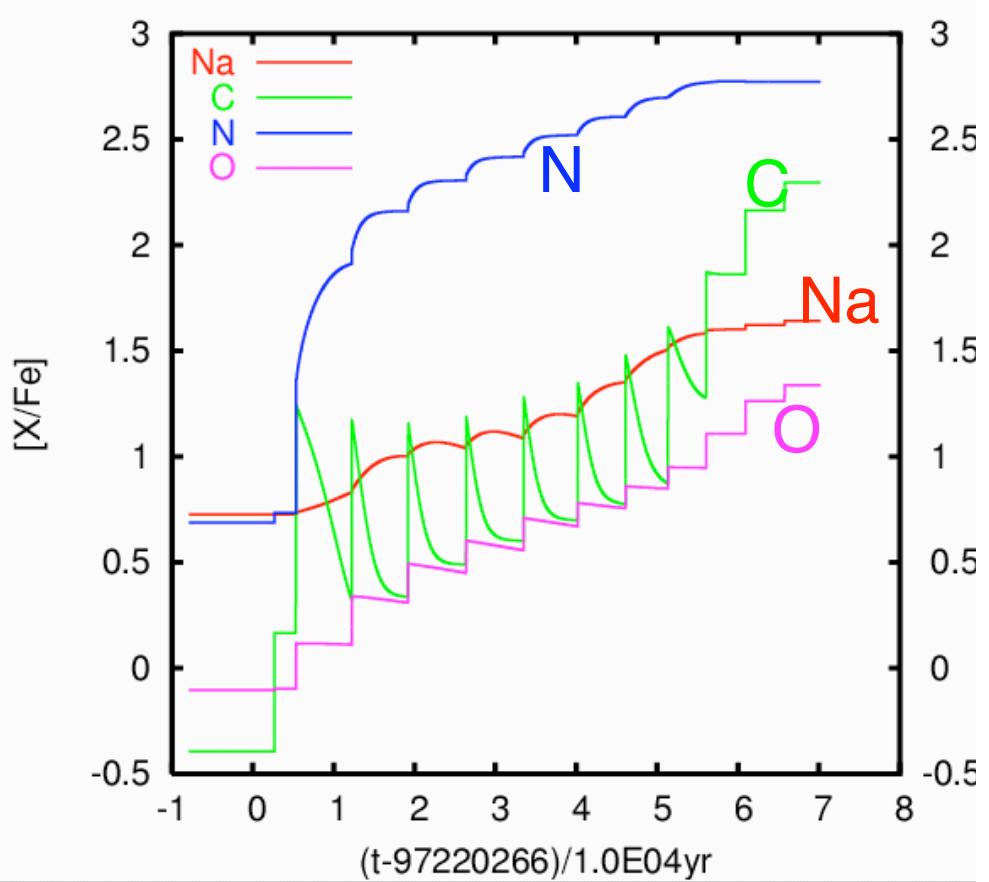
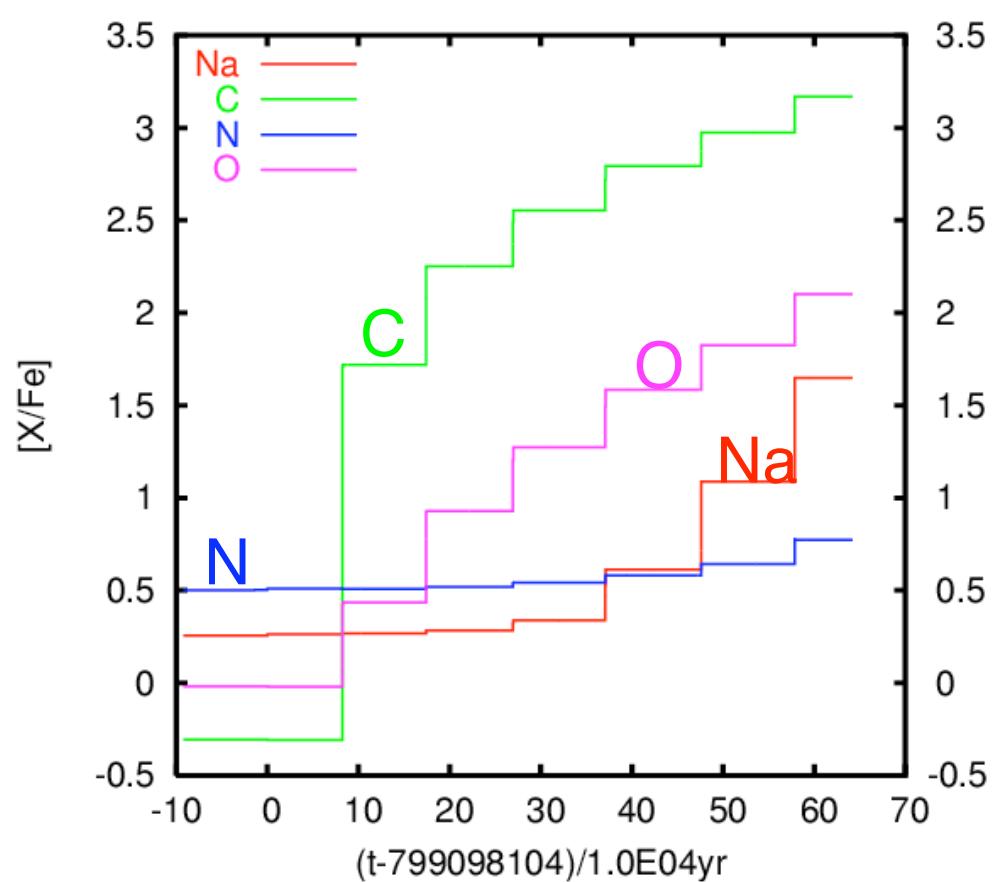
$2M_{\odot}$

ejecta released after:

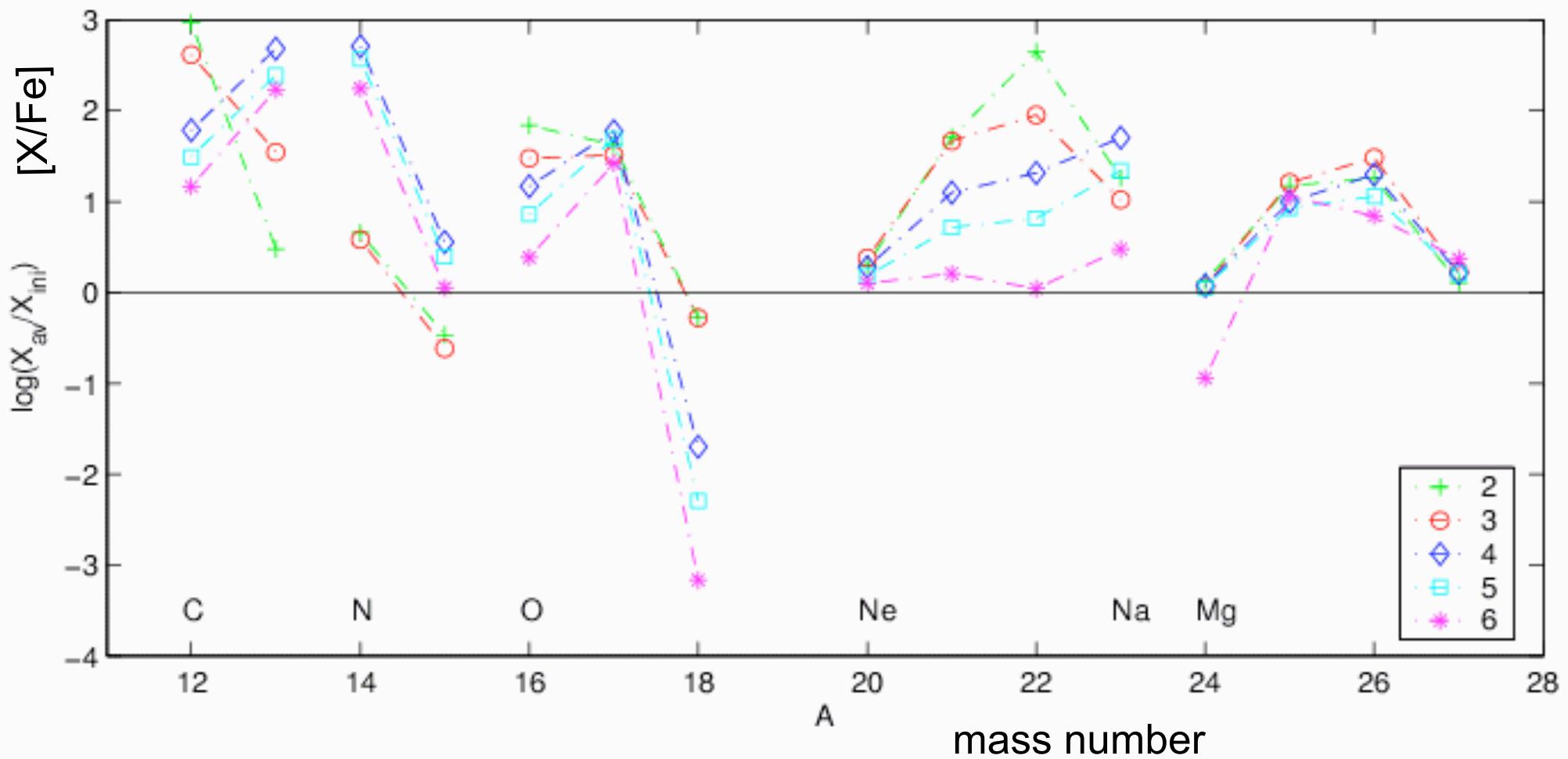
0.8Gyr

$5M_{\odot}$

0.1Gyr

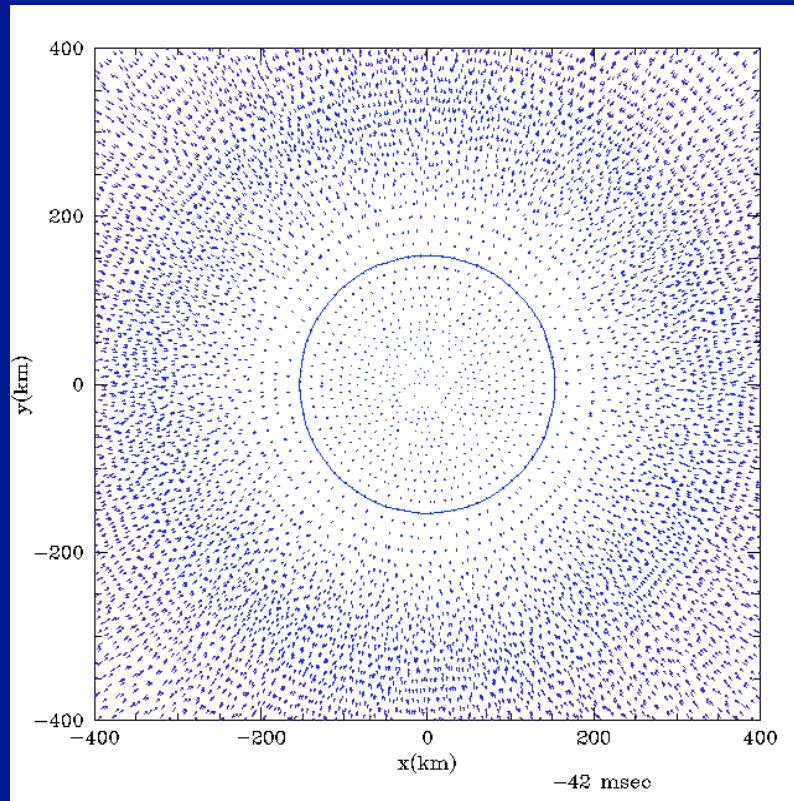


Nuclear production of very low-metallicity AGB stars

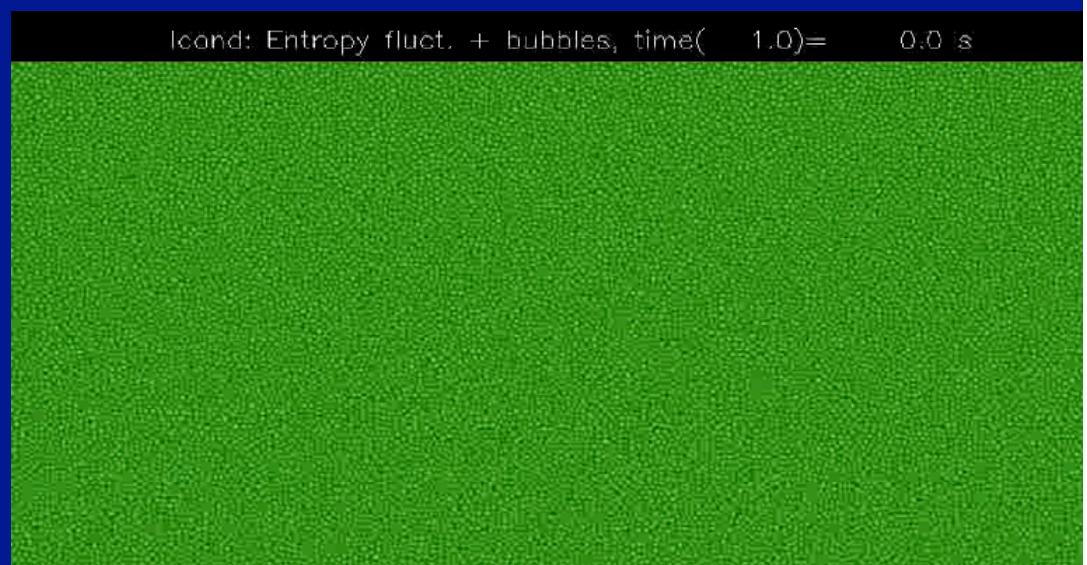


Multi-dimensional simulations:

3D core-collapse supernova simulation with smoothed-particle hydrodynamics



2D He-shell flash convection simulation with grid-hydro code



Outline:

1. What is an element?
2. Elements are made by nucleosynthesis in stars
3. How do we know the abundances of elements in stars:
 - o Stellar spectra
 - o Stellar dust particles in meteorites
4. Calculating the evolution of stars and the making of the elements
5. Scientists paradise: new data, lots of it!
6. Conclusions

New data - new question - new fun!

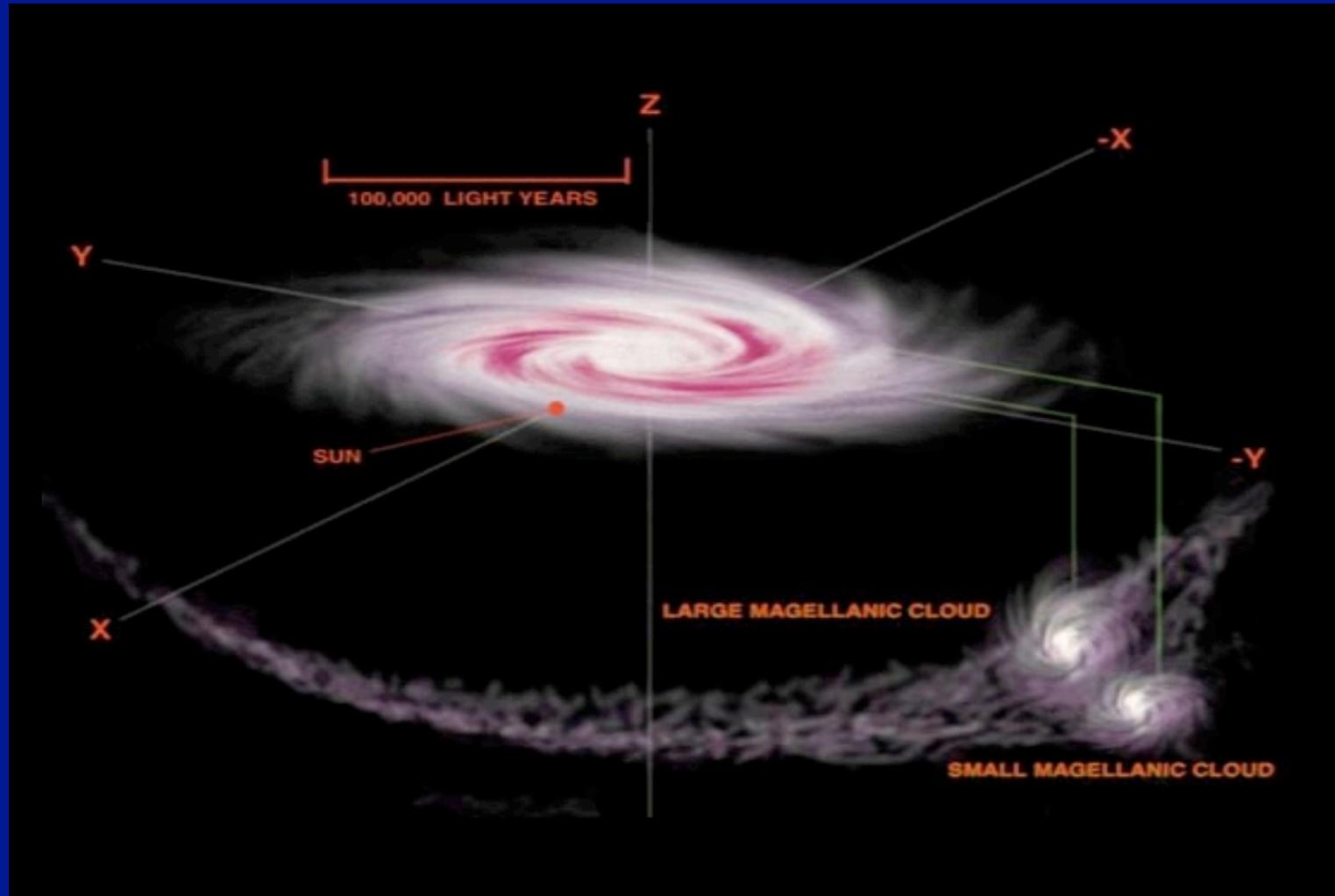
New kid on the block:
Fiber-optics multi-object
spectroscopy

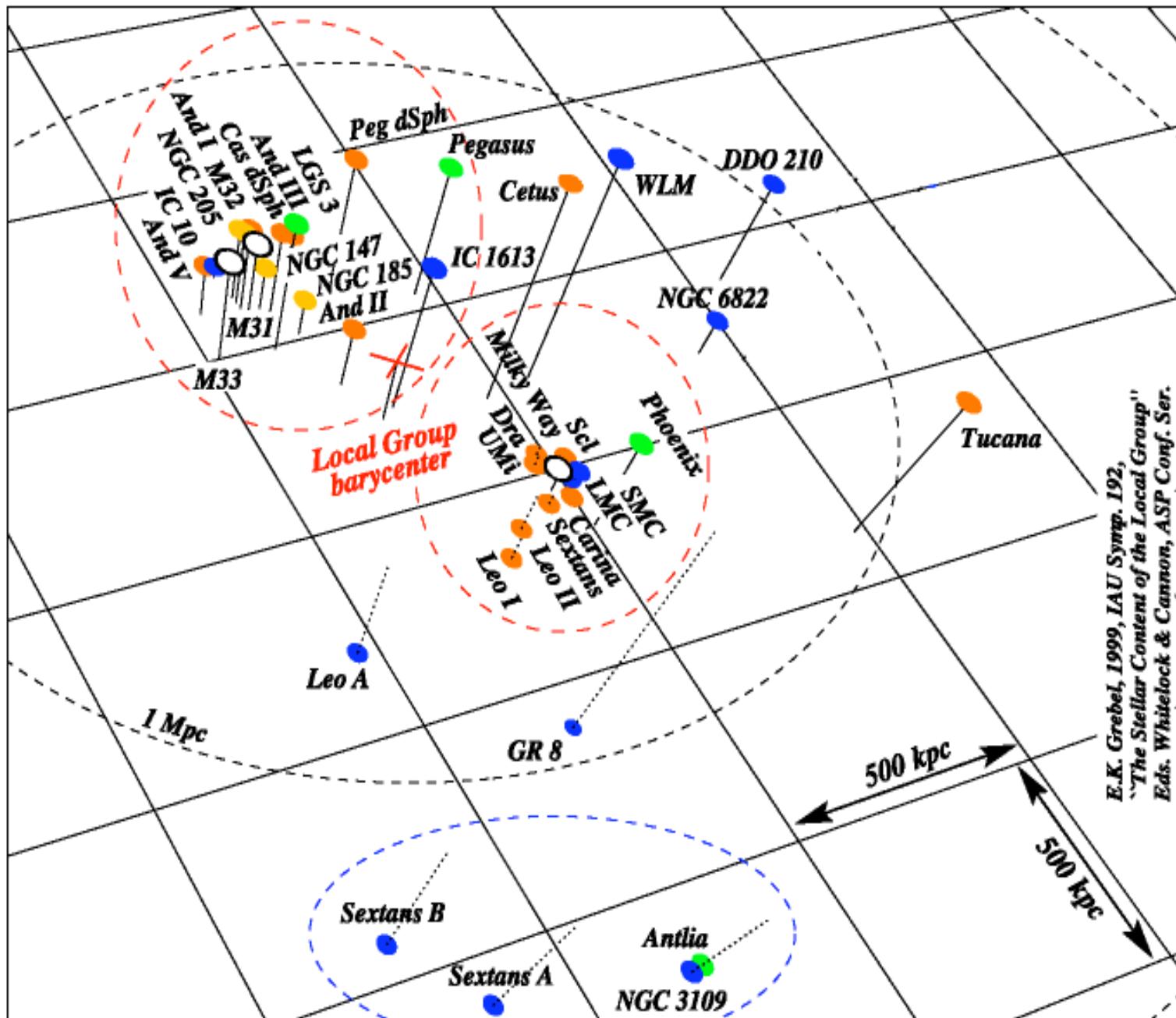
FLAMES at the Very
Large Telescope (VLT,
ESO)

100's to 100's of
spectra in one
shot for stars in
the field of view.



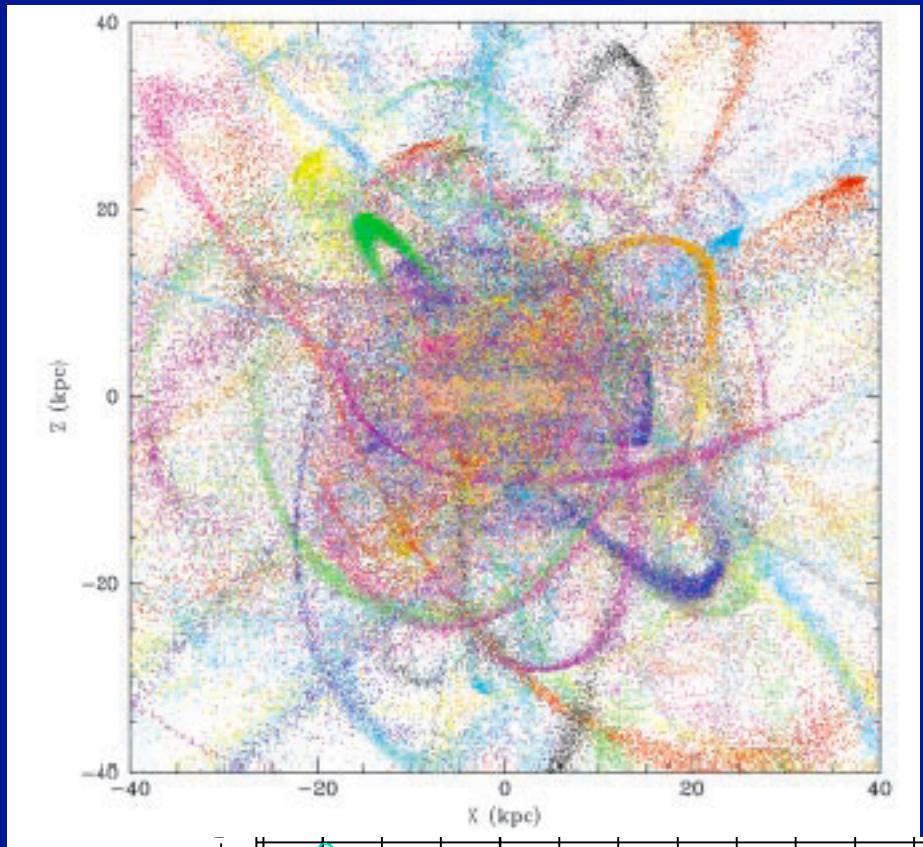
The Milky Way and it's neighbours:





E.K. Grebel, 1999, IAU Symp. 192,
"The Stellar Content of the Local Group"
Eds. Whitelegg & Cannon, ASP Conf. Ser.

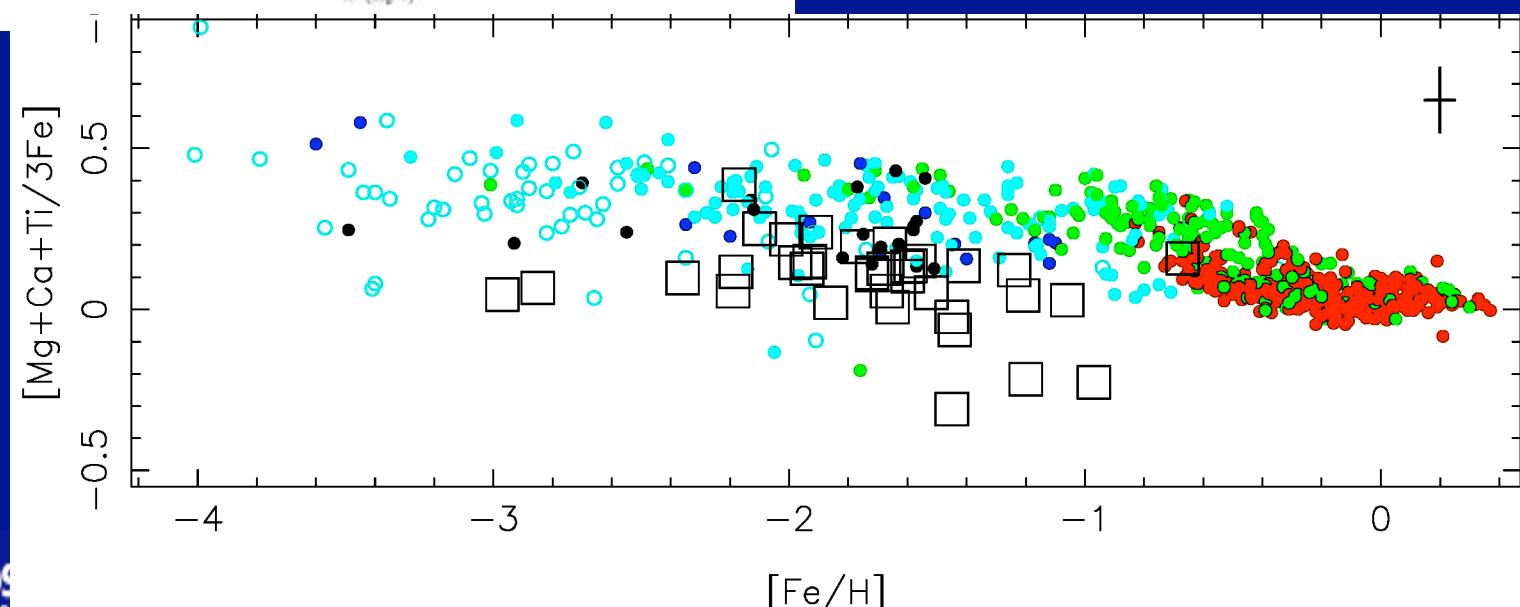
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Framework: Galaxy is the result of a long history of merging events.

<- A simulation of the baryon halo built up through accretion of 100 satellite galaxies.

(Bland-Hawthorn & Freeman, Science 287, 2000)



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